Balance the following redox reactions.

1.
$$H_2SO_3 + I_2 + H_2O \rightarrow H_2SO_4 + HI$$

11.
$$Fe_3O_4 + KMnO_4 + H_2SO_4 \rightarrow Fe_2(SO_4)_3 + K_2SO_4 + MnSO_4 + H_2O_4$$

12.
$$NaCrO_2 + NaOH + H_2O_2 \rightarrow Na_2CrO_4 + H_2O$$

14.
$$Ag_2S + HNO_3 \rightarrow AgNO_3 + NO + S + H_2O$$
 (continued)

(continued)

Answer Key Weston Walch Chemistry Problems

Joacim Rocklöv

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