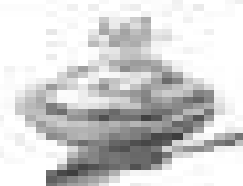


# Atomic Probing Scales of the Atom

Name \_\_\_\_\_  
Date \_\_\_\_\_



## Purpose:

To understand various methods of measuring the size of atoms and molecules.

## Part I: Finding the Size of the Atom

Atoms are the smallest particles of matter which cannot be created or destroyed. They are the building blocks of matter. The size of atoms is measured in terms of radius. The radius of an atom is the distance from the center of the atom to the outermost shell.

**Experiment 1: Estimation of the radius of an atom using the oil drop method.**

Take a small amount of oil and spread it on a clean surface. Measure the area covered by the oil and the thickness of the oil layer. The radius of the atom can be estimated from these measurements.

**Experiment 2: Estimation of the radius of an atom using the X-ray diffraction method.**

Take a small amount of a substance and spread it on a clean surface. Measure the area covered by the substance and the thickness of the substance layer. The radius of the atom can be estimated from these measurements.

**Experiment 3: Estimation of the radius of an atom using the electron diffraction method.**

Take a small amount of a substance and spread it on a clean surface. Measure the area covered by the substance and the thickness of the substance layer. The radius of the atom can be estimated from these measurements.

# Answers Lesson 11 Atomic Pudding

**Arshad Iqbal**



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