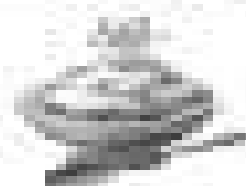


Atomic Probing Structure of the Atom

Name _____
Date _____



Purpose:

To determine which elements are present in a sample of gas by observing the light it emits.

Part I: Making Observations in the Spectroscopy Apparatus

1. Turn on the power supply to the spectroscopy apparatus. The light from the gas tube will be visible through the viewing screen. The light from the gas tube is composed of several discrete lines of color. These lines are called spectral lines.

Observation: The spectrum of the gas tube shows a series of discrete lines of color. The lines are most prominent in the blue and violet regions.

2. Turn on the power supply to the spectroscopy apparatus. The light from the gas tube will be visible through the viewing screen. The light from the gas tube is composed of several discrete lines of color. These lines are called spectral lines.

Observation: The spectrum of the gas tube shows a series of discrete lines of color. The lines are most prominent in the blue and violet regions. The lines are most prominent in the blue and violet regions.

3. Turn on the power supply to the spectroscopy apparatus. The light from the gas tube will be visible through the viewing screen. The light from the gas tube is composed of several discrete lines of color. These lines are called spectral lines.

Observation: The spectrum of the gas tube shows a series of discrete lines of color. The lines are most prominent in the blue and violet regions. The lines are most prominent in the blue and violet regions.

4. Turn on the power supply to the spectroscopy apparatus. The light from the gas tube will be visible through the viewing screen. The light from the gas tube is composed of several discrete lines of color. These lines are called spectral lines.

Answers Lesson 11 Atomic Pudding

RJ Shavelson



Answers Lesson 11 Atomic Pudding:

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