

BIOLOGY
Chapter 2, Lesson 3: Carbon Compounds
Guided Notes

OBJECTIVES

1. Name the elements _____ bonds with to make up life's molecules.
2. Describe the functions of each of the _____ groups of _____

THE CHEMISTRY OF CARBON

- What is so special about carbon that it has its own branch of chemistry called "_____ chemistry"?
- Carbon atoms have four _____ electrons, allowing them to form strong _____ bonds with many other elements and form compounds that living organisms depend on.
 - One carbon atom can bond to another and form _____
 - Carbon-carbon bonds can be _____, _____, or even _____, and carbon chains can even close up on themselves to form _____.

Carbon Structure

Carbon can form single, double, or triple bonds with other atoms. Each line between atoms in a molecular drawing represents one covalent bond.



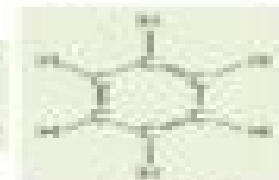
Methane



Acetylene



Butadiene



Benzene

MACROMOLECULES

- Macromolecules: large _____ molecules found in living things. means "_____".
- Produced by a process known as _____, larger compounds are built by joining smaller ones together.
 - _____ (smaller units) are joined together to form _____ (larger compounds).

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University of North Texas Jay Brewster Pepperdine University Laurel Hester South Carolina Governor s School for Science and Mathematics Rather than repeat what is covered in the textbook the Student Study Guide will help students study biology and think like a scientist Introductory chapters on Data Interpretation Looking for Relationships Experimentation and Writing will be illustrated and developed for the student Each text chapter will then be covered with the goal of reinforcing the ideas mentioned in introductory chapters and to tie them to appropriate topics within a chapter

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Table of Contents Answers For Biology Guided Notes

1. Understanding the eBook Answers For Biology Guided Notes
 - The Rise of Digital Reading Answers For Biology Guided Notes
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers For Biology Guided Notes
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For Biology Guided Notes
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For Biology Guided Notes
 - Personalized Recommendations
 - Answers For Biology Guided Notes User Reviews and Ratings
 - Answers For Biology Guided Notes and Bestseller Lists

5. Accessing Answers For Biology Guided Notes Free and Paid eBooks
 - Answers For Biology Guided Notes Public Domain eBooks
 - Answers For Biology Guided Notes eBook Subscription Services
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 - Answers For Biology Guided Notes Enhanced eBook Features
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 - Highlighting and Note-Taking Answers For Biology Guided Notes
 - Interactive Elements Answers For Biology Guided Notes
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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For Biology Guided Notes
9. Balancing eBooks and Physical Books Answers For Biology Guided Notes
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For Biology Guided Notes
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answers For Biology Guided Notes
 - Setting Reading Goals Answers For Biology Guided Notes
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answers For Biology Guided Notes
 - Fact-Checking eBook Content of Answers For Biology Guided Notes
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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