

Calorimetry and Specific Heat Lab

Part 1; Introduction

Purpose- The purpose of this lab is to investigate how a specific heat substance can be determined using a coffee cup calorimeter.

Question- Using a calorimeter, how can we figure out a specific heat?

Hypothesis- If the metal has a low heat, the metal can be a good material for cooking because it will not need as much heat to raise the temperature of the metal in order to cook food.

Part 2; Materials

Thermometer	250 mL beaker	400 mL beaker	Ring stand	Two buret clamps
Water	analytical balance	Test-tube holder	Four test tubes	Hot plate
Lid for polystyrene cup	aluminum, copper, iron, and lead powder	50 mL graduated cylinder	Two polystyrene coffee cups	

Step 1; Gather materials

Step 2; Put together the coffee cup calorimeter.

- I Put one coffee cup in another and then put them in the 250ml beaker.
- I then Cut a hole in the center of the lid to hold our thermometer. Put the lid on the top cup.

Step 3: Measure Masses of the Metals

Answer Sheet For Calorimetry Specific Heat

**National Institute of Standards and
Technology (U.S.)**

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the text box.

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