

Chemistry, Part 1

Module 14 Notes



Big Question

How are metals used in chemical reactions? and produced by chemical reactions?

Big Idea

Using chemical analysis, you can identify the metals of a substance or pure metal and other substances. A metal is often used in manufacturing. It is used in construction as structural analysis with chemical reactions.

Learning Objectives

After completing this module, students will be able to:

- Use the law of conservation of mass to balance chemical reactions.
- Use chemical reactions to describe processes of materials.
- Compare different metals and reactions.
- Describe the properties of different metals.

Module 14 Objectives

Learning Objectives These are the main objectives of chemical reactions.

Fill in the blanks.

1. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (Catalyst)
2. $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$
3. $\text{Ca} + \text{H}_2\text{O} \rightarrow \text{Ca(OH)}_2 + \text{H}_2$

Chemical Reactions and Equations

For the first reaction, the reactants are iron and copper sulfate.

Chemical reactions are the chemical reactions that produce products.

Chemical reactions: $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$

Chemical reaction: $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$

Chemical reactions: $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$

Module 14

Learning Objectives These are the main objectives of chemical reactions.

Fill in the blanks.

1. 1. metals are produced by the reaction of $\text{Fe} + \text{CuSO}_4$
2. The reaction of iron and copper sulfate is the reaction of iron and copper sulfate. The reaction of iron and copper sulfate is the reaction of iron and copper sulfate.

Answers For Chemistry Module 14

**Dr Vijay Sarda, Dr A C Handa, Dr K K
Arora**



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