# UNDERSTANDING THE MOLE AND AVOGADRO'S NUMBER

Norma	Section Date
One mole of anything is a very in- understand just how large a mo- understand how truly large a m	the and Anneadow's Number Salb urge number of things. It is all flowly to be in. This lab convex as a very to one of things in.
tions places. Your goals for this 1. Calculate the volume of the site. You hay measure the solution of several of the ite	
s. Compare the volume of one	place or places that your teacher gives, more of the item to the volume of the
mole of the item solt fit, if to larger, then one mole of the For example: you may be given.	If the volume of the place is larger, one to volume of one male of the familia them will not fit in the place. an ensure so your object and your
theoreum as a phose.  For one mole, use Ausgodni's n	
	-

85	Calculations and the Mol- Item is a
	The volume of this item is
	The volume of one mole of this item is
•	Place is the
	The volume of this place is
•	Will one mole of item 1 fit in this place? Circle one Yes No
	If no, how many of this place will be needed to hold one mole of the item?

A CHEMISTRY LAB
10 PROBLEMS
3 VERSIONS
WORKED ANSWER KEY

## **Answer Key To Understanding The Mole Lab**

**Mocktime Publication** 

#### **Answer Key To Understanding The Mole Lab:**

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