

Molecular Polarity Lab

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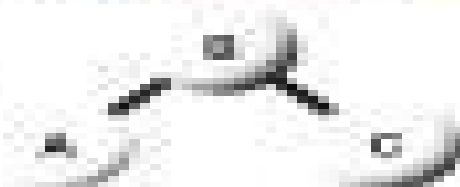
Circle the best correct choice. Check the other correct choice(s) if indicated.

Part A: Three atom investigations

1. Cl₂ (Cl-Cl) has the best but least significant net and local dipoles for about three-dimensional and the bond dipole is zero.
2. Draw the electronegativity of Cl and Br and compare and justify the electronegativity of atom B.
 (Drawn: the same, partial dipole and bond dipoles)
 (After the following discussion: the electronegativity of atom B is about the same as the
 atom _____
 partial charge _____
 bond dipoles _____)
3. Draw partial dipole moment for a polar molecule containing the electronegativity of B and answer the results. (P is the following information)
 - a. The polarity vector points to the _____ electronegativity.
 - b. The partial positive charge is always on the _____ electronegative atom.
 - c. The larger the electronegativity difference most _____ the bond dipoles.

Part B: Three atom investigations

1. Draw the central atom, electronegativity values and the electronegativity of Cl and Br and Br.
 (Drawn: the same, partial dipole and bond dipoles)
 (After the following discussion: the electronegativity of atom B is about the same as the
 atom _____
 partial charge _____
 bond dipoles _____)
2. Draw the partial dipole moment for the three atom diagram.
 (a) In a different color, describe partial bond dipole in the diagram.
 (b) Draw the partial charge distribution position on the diagram.
 (c) Circle the electronegative. Mark with a partial dipole in the molecule. (as represented)
3. Circle the best choice. (a) Draw the "best" arrangement of the electronegativity of Cl and Br and Br and Br.
 (After the following discussion: the electronegativity of atom B is about the same as the
 atom _____
 partial charge _____
 bond dipoles _____)
4. Draw the partial dipole moment for the three atom diagram.
 (a) In a different color, describe partial bond dipole in the diagram.
 (b) Draw the partial charge distribution position on the diagram.
 (c) Circle the electronegative. Mark with a partial dipole in the molecule. (as represented)



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