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## Chem - Covalent Bonding: Questions & Answers

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properties of non metals      ✓ Ans - inability to conduct electricity - absence of ions or delocalised electrons  
low boiling and melting points - generally weak intermolecular bonds, can be soft

molecules      ✓ Ans - separate groups of atoms with a formula, bonded together

intermolecular bonds      ✓ Ans - bonds between molecules (dispersion, dipole-dipole, hydrogen bonding) determines physical properties

intramolecular bonds      ✓ Ans - bonds between atoms in a molecule (covalent, ionic eg.)

covalent bonds      ✓ Ans - non metals share electrons to complete the octet rule, an atom forms as many bonds as it needs eg. carbon needs 4 electrons so forms 4 bonds

why do non metals share electrons not transfer?      ✓ Ans - non metals have a relatively high electronegativity so competing to attract each others electron, shares instead

electron dot diagrams      ✓ Ans - lewis diagram shows valence electrons (bonding pairs) and lone pairs

valence structure      ✓ Ans - each pair of bonding electrons is shown as a line, still represents dots as lone pairs. 1 line=2 electrons

bonding pairs and lone pairs      ✓ Ans - bonding pairs; shared electrons in bond  
lone pairs; non bonding pairs

double covalent bond      ✓ Ans - rigid, two pairs of electrons (4 electrons) are shared, represented as a double line in valence structures eg.  $O_2$

# Answers For Chem Think Covalent Bonding

**Leallyn Burr Clapp**



**Answers For Chem Think Covalent Bonding:**

**Chemistry of the Covalent Bond** Leallyn Burr Clapp, 1957

## Adopting the Melody of Term: An Emotional Symphony within **Answers For Chem Think Covalent Bonding**

In some sort of taken by monitors and the ceaseless chatter of instant connection, the melodic elegance and mental symphony produced by the prepared term frequently disappear in to the background, eclipsed by the persistent sound and disturbances that permeate our lives. Nevertheless, located within the pages of **Answers For Chem Think Covalent Bonding** a marvelous fictional prize brimming with natural feelings, lies an immersive symphony waiting to be embraced. Constructed by an elegant composer of language, that interesting masterpiece conducts visitors on an emotional journey, skillfully unraveling the concealed songs and profound influence resonating within each carefully constructed phrase. Within the depths with this moving analysis, we shall explore the book is central harmonies, analyze its enthralling writing type, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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