

Module 1

ANSWER KEY -The Accounting Process

Activity 1:

1. It is a systematic compilation of a group of accounts.
 - a. Chart of accounts
 - b. Trial balance
 - c. **Ledger**
 - d. Journal
2. It is a list of accounts and their balances.
 - a. Chart of accounts
 - b. **Trial balance**
 - c. Ledger
 - d. Journal
3. Which of the following criteria must be met before an event or item is recorded for accounting purposes?
 - a. The event or item can be measured objectively in financial terms.
 - b. The event or item is relevant and reliable.
 - c. The event affects, or the item meets the definition of, a financial statement element.
 - d. **All of these must be met.**
4. An accounting record into which the essential facts and figures in connection with all transactions are initially recorded is called the
 - a. ledger.
 - b. account.
 - c. trial balance
 - d. **none of these.**
5. A trial balance may prove that debits and credits are equal, but
 - a. an amount could be entered in the wrong account.
 - b. a transaction could have been entered twice.
 - c. a transaction could have been omitted.
 - d. **all of these.**
6. When an item of expense is paid and recorded in advance, it is normally called a(n)
 - a. **prepaid expense.**
 - b. accrued expense.
 - c. estimated expense.
 - d. cash expense.
7. When an item of revenue or expense has been earned or incurred but not yet collected or paid, it is normally called a(n) _____.
 - a. prepaid
 - b. adjusted
 - c. estimated
 - d. **none of these**
8. An unearned revenue can best be described as an amount
 - a. collected and currently matched with expenses.
 - b. **collected but not currently matched with expenses.**
 - c. not collected but currently matched with expenses.
 - d. not collected and not currently matched with expenses.
9. Which of the following is a real (permanent) account?
 - a. Inventory
 - b. Sales
 - c. Accounts Receivable
 - d. **Both Inventory and Accounts Receivable**
10. Reversing entries are

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web solution $v_1 = 250 \text{ l}$ $v_2 = ?$ $t_1 = 273 \text{ K}$ $t_2 = 343 \text{ K}$ we will be applying charles law as pressure is constant here $v_1/t_1 = v_2/t_2$ or $v_2 = v_1 \times t_2 / t_1 = 250 \text{ l} \times 343 \text{ K} / 273 \text{ K} = 311 \text{ l}$ at 90°C a helium sample has a volume of 500 ml determine the temperature at which the volume of the liquid will become 240 ml