

100) Carbon tetrachloride consists of one carbon and four chlorine atoms. How many chlorine atoms are present in 1.00 mole of CCl_4 molecules?

$$\text{one mole of } \text{CCl}_4 \text{ contains } 4 \text{ moles of } \text{Cl} \text{ atoms}$$

$$1 \text{ mole } \text{CCl}_4 \times \frac{4 \text{ moles Cl}}{1 \text{ mole } \text{CCl}_4} = 4 \text{ moles Cl}$$

$$\text{answer} = 4 \text{ moles Cl} \quad \text{100}$$

101) There are 3 monatomic gases in equal molar amounts. What are they?

101) Helium, neon, and argon.

Helium, neon, and argon are the only monatomic gases that exist as gases at room temperature.

Helium, neon, and argon are the only monatomic gases that exist as gases at room temperature.

102) Suggest other three gas law other monatomic gases in equal molar amounts.

Other monatomic gases that exist as gases at room temperature are krypton, xenon, and radon.

103) Sodium chloride is a compound of sodium and chlorine. How many moles of sodium chloride are present in 1.00 mole of NaCl molecules?

103) Sodium chloride is a compound of sodium and chlorine. How many moles of sodium chloride are present in 1.00 mole of NaCl molecules?

There is one sodium atom in each NaCl molecule.

$$\text{answer} = 1 \text{ mole NaCl} \quad \text{103}$$



Answer To F321 Chemistry May Series

Hongru Du



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