

# Chemistry, Part 1

## Module 14 Notes



### Big Question

How are metals used in chemical reactions? and produced by chemical reactions?

### Big Idea

Using chemical analysis, you can identify the metals of a substance or pure elements and their substances. A metal is often used in chemical reactions. It is used in chemical reactions in chemical analysis with chemical reactions.

### Learning Objectives

After completing this module, students will be able to:

- Use the law of conservation of mass to balance chemical reactions.
- Use chemical reactions to determine the composition of mixtures.
- Compare different metals and reactions.
- Describe the properties of different metals.

### Module 14 Big Idea

#### Learning Objectives

How are metals used in chemical reactions?

Fill in the blanks.

1. The law of conservation of mass states that the mass of the reactants is equal to the mass of the products.
2. The law of conservation of mass states that the mass of the reactants is equal to the mass of the products.
3. The law of conservation of mass states that the mass of the reactants is equal to the mass of the products.

### Module 14 Big Idea

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1. The law of conservation of mass states that the mass of the reactants is equal to the mass of the products.
2. The law of conservation of mass states that the mass of the reactants is equal to the mass of the products.

# Answers For Chemistry Module 14

**Sebastian Brünink**



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