

SCH4U Unit 4 - Quiz – Equilibrium

Player: _____

Period: _____

Q1: (10)

Q2: (10)

Q3: (10)

Q4: (10)

Q5: (10)

Write the formulas for:

$$K_{eq} = \frac{[products]}{[reactants]} \quad ; \quad K_{eq} = \frac{P_{products}}{P_{reactants}}$$

$$K_{eq} = \frac{P_{CO_2}^2 P_{H_2O}}{P_{CH_4}^2 P_{O_2}} \quad ; \quad K_{eq} = \frac{P_{CO_2}^2 P_{H_2O}}{P_{CH_4}^2 P_{O_2}}$$

gpt (aq) + h₂o (l) ⇌ gpt(s) + h₂g (g) ; gpt(s) ⇌ h₂g (g)

h₂o (l) ⇌ h₂o (g)

Question 1:

1. Hydrogen sulfide, H₂S, is a slightly poisonous gas. At 1000°C, an equilibrium mixture contains 0.020 moles of hydrogen, H₂, 0.010 moles of hydrogen sulfide, and an undetermined amount of sulfur, S₂. The reaction is:



If the value of K is 1.000, and if 0.020 moles of H₂ are present at equilibrium,

calculate:

a. [S₂]

b. [H₂S] at equilibrium

c. [S₂] at 1000°C

d. Determine [S₂] at equilibrium

e. Answer:

and what is the best chemical formula used as a product in the above?

f. Answer:

	Q1: (10)	Q2: (10)	Q3: (10)
Q1: (10)			
Q2: (10)			
Q3: (10)			

1. The following equilibrium:

Answers For Sch4u Unit 4

LP Steffe



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