

Answer following



- 1. Substrate and temperature: **200, 250, 300**
- 2. A graph of product formed: **product formed, 200, 250, 300**
- 3. A graph of initial rate: **initial rate, 200, 250, 300**

Examine all graphs of rate & time at temperature change.
 Rate of reaction is proportional to the rate of change.

- 1. increasing weight: **10, 20, 30, 40, 50**
- 2. increasing frequency: **10, 20, 30, 40, 50**
- 3. increasing volume: **10, 20, 30, 40, 50**
- 4. increasing pressure: **10, 20, 30, 40, 50**



Example 10.1

Graphs of rate of reaction vs. time at different temperatures are shown below. The graphs are given below.



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Answers Of 11 3 Prentice Hall Geometry

A Gutmann



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