

Chemical Bonding

Ionic Bonding:

- electrostatic force of attraction between ~~positively~~ oppositely charged ions.
- It involves transfer of e^- (between a metal and a non-metal)

Covalent Bonding:

- electrostatic attraction between the nuclei of the bonded atoms and the shared pair of e^- between them
- It involves sharing of e^- (between 2 non-metals.)

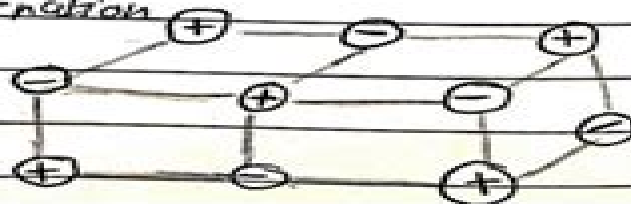
metallic Bonding:

- electrostatic force of attraction between layers of cations and sea of delocalised e^- .

Structure

Ionic compound

Ionic lattice: regular arrangement of $(+)$ / $(-)$ ions showing an alternation between them.



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