#### Kinetics Worksheet 1

3	For the reaction:	A + 2B	$\rightarrow$ 3 C $+$ D	Ġ	
-	the following data were obtained,				
	271 6101	EA.3	COL		

STREET, SQUARE	AS ANDARAM DAGAS.		
mum	[A]	[8]	rate M-s-1
1	0.16	0.36	2.67
2	0.16	0.09	0.67
3	0.32	0.24	7.11
4	0.08	0.09	0.168
5	0.24	0.12	2.0
6	0.02	1.35	7

$$\frac{(0.16)^{1}}{(0.08)^{1}} = \frac{0.67ms^{-1}}{0.168ms^{-1}}$$

$$(2)^{1} = 4$$

$$1 = 4$$

$$1 = 4$$

$$1 = 8$$

$$1 = 8$$

$$1 = 8$$

$$1 = 8$$

$$1 = 8$$

$$1 = 8$$

$$1 = 8$$

$$1 = 8$$

$$1 = 8$$

$$1 = 8$$

$$1 = 8$$

$$1 = 8$$

$$1 = 8$$

(C) Calculate the rate of the reaction when 
$$[A] = 0.02 \text{ M}$$
 and  $[B] = 1.35 \text{ M}$ .

#### (D) Assume that the reaction goes to completion. Under the conditions specified in the second run, what would the final [C] be?

$$A + 2B \rightarrow 3C + D$$
  
 $16A_{1\overline{A}} = 0.32B$   
1. 1. 2.09B. 14 = 0.045A





# **Answers Kinetics Worksheet 1**

Jonathan Birkin

#### **Answers Kinetics Worksheet 1:**

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