Name: Key

For each reaction:

- * Assign Oxidation Numbers
- * Write an Oxidation and Reduction Half Reaction
- *Circle the Oxidizing Agent and Underline the Reducing Agent.

1) Fe⁻³ + 5n → Fe⁻² + 5n⁻²

Oxidation: Snº -> Snº + 2e

Reduction: (Fe') . Is' -> Fe'

2) AlF, * Bo -> Al * BoF.

Oxidation: Bot + Bort + 2e-

Reduction: (Al-3) + 3e - Al-

3) H₂O₂ + H₂SO₂ → H₂O + H₂SO₂

Oxidation: 5-1 + 5-1 + 2e

Reduction: (O1) + Ie: + O-

4) $H_1SO_1 + KMnO_4 \rightarrow MnSO_4 + K_1SO_4 + H_2O_5$

Oxidation: 5" + 5" + 2e

Reduction: (Mn-) + 5e - Mn-2

5) K2Cr2O7 + H2C2O4 + H2SO4 + K2SO4 + Cr2(SO4)3 + CO2 + H2O

Oxidation: C1 -> C-1 + 1e

Reduction: (Cris) . 3c -> Cris

Analyzing Oxidation Reduction Reactions Answers Sheet

Charlotte W. Pratt, Kathleen Cornely

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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Analyzing Oxidation Reduction Reactions Answers Sheet**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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