

5. When neutral atoms **gain** one or more electrons, _____ are formed.
- hadrons
 - cations
 - fermions
 - deuterions
 - anions**
6. Which two of the following represent a **pair of isotopes**?
- | | | | | |
|----------------------|----------------------|----------------------|----------------------|----------------------|
| $^{31}_{12}\text{P}$ | $^{32}_{16}\text{P}$ | $^{32}_{15}\text{P}$ | $^{32}_{16}\text{S}$ | $^{33}_{16}\text{S}$ |
| I. | II. | III. | IV. | V. |
- I and IV
 - III and IV
 - I and III**
 - I and II
 - II and V
7. Which of the following consists of a pair of **allotropes**?
- air and oxygen
 - phosgene, $\text{O}=\text{C}-(\text{Cl})_2$ and phosphine, PH_3
 - red phosphorus and white phosphorus**
 - carbon tetrachloride and carbon tetrabromide
 - yellow sulfur and yellow phosphorus
8. An element that has 24 protons, 24 electrons, and 28 neutrons has a molar mass of _____.
- 76 amu.
 - 56 amu.
 - 52 amu.**
 - 48 amu.
 - 84 amu.
9. Three elements that are likely to have **similar** chemical and physical properties are:
- Ca, Co, Cr
 - H, He, Ne
 - Pb, Bi, Po
 - B, C, Si
 - Na, K, Li**
10. Which formula and name combination is **incorrect**?
- SF_6 and sulfur hexafluoride
 - N_2O_5 and dinitrogen tetraoxide**
 - P_4O_{10} and tetraphosphorus decaoxide
 - AsBr_5 and arsenic pentabromide
 - SeF_4 and selenium tetrafluoride

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1. Understanding the eBook Aau Chem 101 Exam Past Questions For 2014
 - The Rise of Digital Reading Aau Chem 101 Exam Past Questions For 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Aau Chem 101 Exam Past Questions For 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Aau Chem 101 Exam Past Questions For 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Aau Chem 101 Exam Past Questions For 2014
 - Personalized Recommendations
 - Aau Chem 101 Exam Past Questions For 2014 User Reviews and Ratings
 - Aau Chem 101 Exam Past Questions For 2014 and Bestseller Lists
5. Accessing Aau Chem 101 Exam Past Questions For 2014 Free and Paid eBooks
 - Aau Chem 101 Exam Past Questions For 2014 Public Domain eBooks

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- Aau Chem 101 Exam Past Questions For 2014 Budget-Friendly Options
- 6. Navigating Aau Chem 101 Exam Past Questions For 2014 eBook Formats
 - ePub, PDF, MOBI, and More
 - Aau Chem 101 Exam Past Questions For 2014 Compatibility with Devices
 - Aau Chem 101 Exam Past Questions For 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Aau Chem 101 Exam Past Questions For 2014
 - Highlighting and Note-Taking Aau Chem 101 Exam Past Questions For 2014
 - Interactive Elements Aau Chem 101 Exam Past Questions For 2014
- 8. Staying Engaged with Aau Chem 101 Exam Past Questions For 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Aau Chem 101 Exam Past Questions For 2014
- 9. Balancing eBooks and Physical Books Aau Chem 101 Exam Past Questions For 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Aau Chem 101 Exam Past Questions For 2014
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Aau Chem 101 Exam Past Questions For 2014
 - Setting Reading Goals Aau Chem 101 Exam Past Questions For 2014
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Aau Chem 101 Exam Past Questions For 2014
 - Fact-Checking eBook Content of Aau Chem 101 Exam Past Questions For 2014
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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