

2. (a) An aqueous solution of a salt was added to excess sodium trioxocarbonate(IV) solution in a test tube. There was effervescence, the test tube became warm and a white precipitate was observed. State three inferences that could be drawn from these observations.

- (b) Consider the following salts:

NH_4Cl ; PbSO_4 ; NaHCO_3 ; and $\text{Cu}(\text{NO}_3)_2$

Select from the list, the salt(s) which:

- (i) would not readily dissolve in water;
-
- (ii) produce(s) effervescence with dilute mineral acids;
-
- (iii) decompose(s) on heating;
-
- (iv) dissolve(s) in water to form an alkaline solution;
-
- (v) sublime(s) on heating.
-

Answers To Waec 2015 Chemistry Practical Alternative A

G Thomas



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