

STATISTICS
ECONOMICS

Tenth Edition

Elementary Statistics

A STEP BY STEP APPROACH

ALLAN G. BLUMAN

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oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^{-}$ reduction $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$ 24 3 $\text{Ag} + \text{Ni}^{3+} \rightarrow \text{Ni} + 3\text{Ag}^{+}$ oxidation reduction 25 $\text{Cu}^{2+} + \text{Pb} \rightarrow \text{Pb}^{2+} + \text{Cu}$ oxidation reduction 26 O_2

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magnitude of change then identify the element oxidized and reduced as well as the oxidizing agent and reducing agent a i2o5
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reduction reactions or redox reactions in this unit you will examine the oxidation reduction process and use the oxidation
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