

ANSWER KEY

Vocabulary Workshop Level E Unit 1 Test

adulterated	ambidextrous	augment	benefit
deceit	dour	fortitude	gaped
gibes	guise	insidious	intimation
opulent	pliable	reiterate	stolid
imitate	unkempt	verbatim	warily

◆◆◆◆◆

adj. 1. From the **stolid** expression on his face, I could not tell whether he was interested in taking on the job or not.

n. 2. It requires tremendous **fortitude** to stand up to peer pressure and do what you know is right not what is popular.

adj. 3. The **opulent** setting for the charity ball made some of the kids from the poorer neighborhoods stare in amazement.

adj. 4. The writer was **ambidextrous**; he could write any kind of article or story on demand, which made him very valuable to his employers.

adv. 5. For some children, a useful discipline technique is to have them repeat your instructions back to you **verbatim**.

vb. 6. The agency charged that the manufacturer had willfully **adulterated** the product with cheap fillers and impure additives.

adv. 7. Uncertain whether the tranquilizer dart had fully incapacitated the lion, the gamekeepers approached the fallen animal **cautiously**.

n. 8. The undercover detective adopted the **guise** of a thief in order to infiltrate the ranks of the crime ring.

adj. 9. The **insidious** power of the secret police grew daily; the citizens never aware of their danger.

adj. 10. The boy's **unkempt** appearance caused his teachers to wonder whether he was being properly cared for at home.

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VB = verb

ADJ = adjective

ADV = adverb

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S Ashworth



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is 0 oxidation ni s ni a 2 a q 2 e now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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web aug 29 2023 some points to remember when balancing redox reactions the equation is separated into two half equations one for oxidation and one for reduction the equation is balanced by adjusting coefficients and adding H_2O H^+ and e^- in this order balance the atoms in the equation apart from o and h

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web in the redox equation below that occurs in our stomach which of the following pairs identifies the reducing and oxidizing agents respectively $\text{C}_6\text{H}_8\text{O}_6 + 2\text{H}^+ + 2\text{NO}_2^- \rightarrow \text{C}_6\text{H}_6\text{O}_6 + 2\text{H}_2\text{O} + 2\text{NO}$ choose 1 answer $\text{C}_6\text{H}_8\text{O}_6$ and NO_2^- a $\text{C}_6\text{H}_8\text{O}_6$ and NO_2 b NO_2 and $\text{C}_6\text{H}_8\text{O}_6$ c $\text{C}_6\text{H}_8\text{O}_6$ and H^+ d H^+ and $\text{C}_6\text{H}_8\text{O}_6$