

Analysis of an Antacid

By: Jonathan Pilafas (performed on 10/15/2018)

Purpose: The objective of this lab is to compare different generic and name brand antacids to determine if they are equally effective in neutralizing acid and if they are comparably cost effective.

References:

1. "Laboratory Experiments for Chemistry 121" - Harper College
2. <http://dept.hamercollage.edu/chemistry/msds1/Phenolphthalein%20ITBaker.pdf>
3. <http://dept.hamercollage.edu/chemistry/msds1/Hydrochloric%20acid%20Fisher.pdf>
4. <http://dept.hamercollage.edu/chemistry/msds1/Sodium%20hydroxide%20Fisher.pdf>
5. <http://dept.hamercollage.edu/chemistry/msds1/Calcium%20carbonate%20Fisher.pdf>

Reagents:

Name	Formula	Molecular Weight	Concentration	Hazards
Hydrochloric Acid	HCl	36.46 g/mol	36-38%	Health - 3 Fire - 0 Reactivity - 1
Sodium Hydroxide	NaOH	40.00 g/mol	50%	Health - 3 Fire - 0 Reactivity - 1
Phenolphthalein	$C_{20}H_{14}O_4$	318.312 g/mol	N/A	Health - 2 Fire - 1 Reactivity - 0
Tums	$CaCO_3$	Active Ingredient in Antacid	N/A	Health - 2 Fire - 0 Reactivity - 0

Antacid Analysis Lab Answers

David Baud

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

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