

AP Bio Chapter 06 Group Notes: Metabolism - Energy and Enzymes

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Team Members:

HSR&T (Highly Suggested Reading and Thinking) **This is independent work that you can do as a (GSN) team/individually - either way, ALL members of the team should do the reading/thinking....you will be assessed over this information.**

Big Idea 4: Interdependent Relationships ~ "Enzyme Inhibitors Can Spell Death" (p. 108)

- How do the following specifically impact organisms? ---also where have they been **discovered** or **used by people**?
 - Cyanide gas/MPTP:
 - Sarin:
 - Warfarin:
- Give an example of how our understanding of warfarin has had positive consequences:

6.1 Cells and the Flow of Energy 3.4

Energy: the ability to do -work- or bring about a change All energy - dependent upon the sun ultimately! - Kinetic energy: anything -moving- has it - Potential energy: Stored energy - Chemical energy: form of potential energy that is stored in -molecules- - dependent upon its -bonds-	
Laws of Thermodynamics: - Basics: the study of energy transformations that occur in matter - First Law: - also called: -Law of Conservation of Energy- - states: energy cannot be created or destroyed, but it can be transferred or transformed. - example: solar energy used to form carbohydrates by autotrophs - Second Law: - states: every energy transfer or transformation increases the entropy, or the amount of -disorder/randomness- in the universe - example: heat generated by you working out, no transfer of energy is 100% efficient Summary: Can't create energy - only -convert-, at every conversion, some of the energy will dissipate as -heat- and become less useful.	Autotrophs= Primary producers that can create organic energy using non-organic materials like carbon-dioxide or solar energy (Plants are a form of autotrophs)
Entropy: - Defined: Relative amount of -disorganization- - Examples: Glucose breaking down into carbon dioxide and water - Summary: - energy conversions result in heat, therefore, the entropy of the universe is	Heat = Increased entropy in the universe (increased amount of disorganization)

Ap Biology Chapter 6 Notes

Mark Anestis, Kellie Ploeger Cox



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Table of Contents Ap Biology Chapter 6 Notes

1. Understanding the eBook Ap Biology Chapter 6 Notes
 - The Rise of Digital Reading Ap Biology Chapter 6 Notes
 - Advantages of eBooks Over Traditional Books
2. Identifying Ap Biology Chapter 6 Notes
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ap Biology Chapter 6 Notes
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ap Biology Chapter 6 Notes
 - Personalized Recommendations

- Ap Biology Chapter 6 Notes User Reviews and Ratings
- Ap Biology Chapter 6 Notes and Bestseller Lists
- 5. Accessing Ap Biology Chapter 6 Notes Free and Paid eBooks
 - Ap Biology Chapter 6 Notes Public Domain eBooks
 - Ap Biology Chapter 6 Notes eBook Subscription Services
 - Ap Biology Chapter 6 Notes Budget-Friendly Options
- 6. Navigating Ap Biology Chapter 6 Notes eBook Formats
 - ePub, PDF, MOBI, and More
 - Ap Biology Chapter 6 Notes Compatibility with Devices
 - Ap Biology Chapter 6 Notes Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ap Biology Chapter 6 Notes
 - Highlighting and Note-Taking Ap Biology Chapter 6 Notes
 - Interactive Elements Ap Biology Chapter 6 Notes
- 8. Staying Engaged with Ap Biology Chapter 6 Notes
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ap Biology Chapter 6 Notes
- 9. Balancing eBooks and Physical Books Ap Biology Chapter 6 Notes
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ap Biology Chapter 6 Notes
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ap Biology Chapter 6 Notes
 - Setting Reading Goals Ap Biology Chapter 6 Notes
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ap Biology Chapter 6 Notes
 - Fact-Checking eBook Content of Ap Biology Chapter 6 Notes

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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