chemical analysis

Period:

Chapter 4: Carbon and the Molecular Diversity of Life time atmosphere carrietaned Shark 5 discharged to a majore of of hydrogen gas Concept 4.1 Organic chemistry is the study of carbon compounds (H,), mirrote (CH,), and IMPROVING INCIDENTIANO "Atmosphere" 1. Study this figure of Stanley Miller's experiment to simulate conditions thought to have existed on the Winter vegeor-Electrode early Earth. Explain the elements of this experiment. using arrows to indicate what occurs in various parts of the apparatus. Comdonwer Cooled water containing - Chestieti onganie. a condensor cooled molecules the altmosphere raining water and any dissoved ent water wenterpresent to FIRSK WOS money when mented. C. disserved into the Viopes entere **PLOUSING** TWE TO TWO ST Sample for OS material cycled

Name

What was collected in the sample for chemical analysis? What was concluded from the results of this collected. experiment? There were a variety of organic molecules matare common in drganisms (and drocar born, amino acids, CH20, CHU and more)

concluded that organic molecules may have been synthesized abiologily on the earth

Concept 4.2 Carbon atoms can form diverse molecules by bonding to four other atoms

- Make an electron distribution diagram of carbon. It is essential that you know the answers to these questions:
 - a. How many valence electrons does carbon have? 4!!!
 - b. How many bonds can carbon form?
 - b. What type of bonds does it form with other elements?

covalent bonds

- Carbon chains form skeletons. List here the types of skeletons that can be formed.
 - 3) presence of mings in length

2.) Branching 4.) double bond position

What is a hydrocarbon? Name two. Are hydrocarbons hydrophobic or hydrophilic?

an organic molecule consisting only of carbon &

HUGTOP MODIC

enrough the aparents

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JA Banks

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