Name:

Let's see about reviewing the concepts we've learned in this unit!

- Consider this mass spectrum for the element boron to the right.
 - a. How many naturally occurring isotopes does boron have?
 - b. Calculate the atomic mass of boron— check your answer with the atomic mass on the periodic table. Show Your work!



- c. How does boron's atomic mass compare with the mass numbers of its isotopes?
- 2.) A nitrate of lead is found to contain 45.52% lead. What is the formula for this salt?

- 3.) A hydrocarbon with a molar mass of approximately 85, is found to contain 85.7% carbon by mass.
 - a.) calculate its empirical formula:
 - b.) Determine its molecular formula:
- Calculate number of oxygen atoms in 20.00g of zinc phosphate.

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