

NOMENCLATURE REVIEW

Mineral Compounds, Ionic Compounds, & Acids ANSWER KEY

1. <u>potassium sulfate</u>	21. <u>barium</u>
2. <u>potassium hydrogen sulfate</u>	22. <u>nickel(II) oxide</u>
3. <u>phosphorus pentachloride</u>	23. <u>calcium</u>
4. <u>calcium carbonate</u>	24. <u>NO₂</u>
5. <u>calcium carbonate</u>	25. <u>LiOH</u>
6. <u>phosphoric acid</u>	26. <u>calcium</u>
7. <u>calcium dichloride</u>	27. <u>tin(IV)</u>
8. <u>calcium carbonate</u>	28. <u>NO₂</u>
9. <u>calcium chloride</u>	29. <u>tin(IV)</u>
10. <u>calcium(II) chloride</u>	30. <u>LiOH</u>
11. <u>hydrochloric acid</u>	31. <u>lead</u>
12. <u>ammonium carbonate</u>	32. <u>Na₂</u>
13. <u>calcium carbonate</u>	33. <u>hydro</u>
14. <u>calcium sulfate</u>	34. <u>Na₂</u>
15. <u>calcium phosphate</u>	35. <u>hydro</u>
16. <u>calcium sulfate</u>	36. <u>Na₂</u>
17. <u>hydrochloric acid</u>	37. <u>tin(IV)</u>
18. <u>ammonium chloride</u>	38. <u>NO₂</u>
19. <u>calcium sulfate</u>	39. <u>hydro</u>
20. <u>calcium carbonate</u>	40. <u>Na₂</u>
21. <u>calcium carbonate</u>	41. <u>Na₂</u>
22. <u>calcium carbonate</u>	42. <u>tin(IV)</u>
23. <u>calcium carbonate</u>	43. <u>Na₂</u>
24. <u>calcium sulfate</u>	44. <u>tin(IV)</u>
25. <u>calcium(II) chloride</u>	45. <u>Na₂</u>
26. <u>calcium sulfate</u>	46. <u>tin(IV)</u>
27. <u>calcium carbonate</u>	47. <u>Na₂</u>
28. <u>calcium carbonate</u>	48. <u>NO₂</u>
29. <u>calcium sulfate</u>	49. <u>tin(IV)</u>
30. <u>calcium(II) chloride</u>	50. <u>Na₂</u>

Answers To The Nomenclature Review Sheet

JG Myers



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