

1999 Advanced Placement Chemistry Exam

Section I: Multiple Choice

Time — 1 hour and 30 minutes

NO CALCULATORS MAY BE USED WITH SECTION I.

Note: For all questions, assume that the temperature is 298 K, the pressure is 1.00 atmosphere, and solutions are aqueous unless otherwise specified.

Throughout the test the following symbols have the definitions specified unless otherwise noted.

T = temperature
 P = pressure
 V = volume
 S = entropy
 H = enthalpy
 G = free energy
 R = molar gas constant
 n = number of moles
 M = molar

m = molal
 L, mL = liter(s), milliliter(s)
 g = gram(s)
 nm = nanometer(s)
 atm = atmosphere(s)
 J, kJ = joule(s), kilojoule(s)
 V = volt(s)
 mol = mole(s)

Part A

Directions: Each set of lettered choices below refers to the numbered statements immediately following it. Select the one lettered choice that best fits each statement and then fill in the corresponding oval on the answer sheet. A choice may be used once, more than once, or not at all in each set.

Questions 1–4 refer to the following types of energy.

- (A) Activation energy
- (B) Free energy
- (C) Ionization energy
- (D) Kinetic energy
- (E) Lattice energy

1. The energy required to convert a ground-state atom in the gas phase to a gaseous positive ion
2. The energy change that occurs in the conversion of an ionic solid to widely separated gaseous ions
3. The energy in a chemical or physical change that is available to do useful work
4. The energy required to form the transition state in a chemical reaction

Questions 5–8 refer to atoms for which the occupied atomic orbitals are shown below.

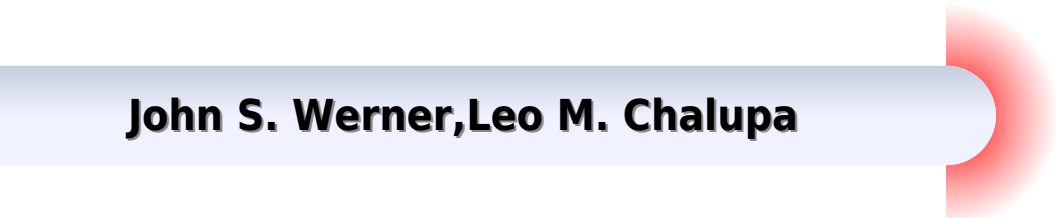
- (A) $1s \underline{\hspace{1cm}} 2s \underline{\hspace{1cm}}$
- (B) $1s \underline{\hspace{1cm}} 2s \underline{\hspace{1cm}}$
- (C) $1s \underline{\hspace{1cm}} 2s \underline{\hspace{1cm}} 2p \underline{\hspace{1cm}} \underline{\hspace{1cm}} \underline{\hspace{1cm}}$
- (D) $1s \underline{\hspace{1cm}} 2s \underline{\hspace{1cm}} 2p \underline{\hspace{1cm}} \underline{\hspace{1cm}} \underline{\hspace{1cm}}$
- (E) $[Ar] 4s \underline{\hspace{1cm}} 3d \underline{\hspace{1cm}} \underline{\hspace{1cm}} \underline{\hspace{1cm}} \underline{\hspace{1cm}}$

5. Represents an atom that is chemically unreactive
6. Represents an atom in an excited state
7. Represents an atom that has four valence electrons
8. Represents an atom of a transition metal

Questions 9–12 refer to aqueous solutions containing 1:1 mole ratios of the following pairs of substances. Assume all concentrations are 1 M .

Ap Chemistry Exam Multiple Choice Answers 2008

John S. Werner, Leo M. Chalupa



Ap Chemistry Exam Multiple Choice Answers 2008:

Guide to College Majors 2008 Princeton Review (Firm), 2008-02-05 Fully revised and updated for 2008 this resource includes the most current information available on more than 350 of the most popular majors Each profile includes an overview of the major career options and earning potential sample college curricula and more **AP Chemistry For Dummies** Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag, 2008-11-13 A practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam Gearing up for the AP Chemistry exam AP Chemistry For Dummies is packed with all the resources and help you need to do your very best Focused on the chemistry concepts and problems the College Board wants you to know this AP Chemistry study guide gives you winning test taking tips multiple choice strategies and topic guidelines as well as great advice on optimizing your study time and hitting the top of your game on test day This user friendly guide helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out of your AP course You ll get help understanding atomic structure and bonding grasping atomic geometry understanding how colliding particles produce states and so much more To provide students with hands on experience AP chemistry courses include extensive labwork as part of the standard curriculum This is why the book dedicates a chapter to providing a brief review of common laboratory equipment and techniques and another to a complete survey of recommended AP chemistry experiments Two full length practice exams help you build your confidence get comfortable with test formats identify your strengths and weaknesses and focus your studies You ll discover how to Create and follow a pretest plan Understand everything you must know about the exam Develop a multiple choice strategy Figure out displacement combustion and acid base reactions Get familiar with stoichiometry Describe patterns and predict properties Get a handle on organic chemistry nomenclature Know your way around laboratory concepts tasks equipment and safety Analyze laboratory data Use practice exams to maximize your score Additionally you ll have a chance to brush up on the math skills that will help you on the exam learn the critical types of chemistry problems and become familiar with the annoying exceptions to chemistry rules Get your own copy of AP Chemistry For Dummies to build your confidence and test taking know how so you can ace that exam *Developing and Validating Test Items* Thomas M. Haladyna, Michael C. Rodriguez, 2013-07-18 Since test items are the building blocks of any test learning how to develop and validate test items has always been critical to the teaching learning process As they grow in importance and use testing programs increasingly supplement the use of selected response multiple choice items with constructed response formats This trend is expected to continue As a result a new item writing book is needed one that provides comprehensive coverage of both types of items and of the validity theory underlying them This book is an outgrowth of the author s previous book *Developing and Validating Multiple Choice Test Items* 3e Haladyna 2004 That book achieved distinction as the leading source of guidance on creating and validating selected response test items Like its predecessor the content of this new book is based on both an extensive review of the literature and on its

author's long experience in the testing field. It is very timely in this era of burgeoning testing programs especially when these items are delivered in a computer-based environment. Key features include Comprehensive and Flexible. No other book so thoroughly covers the field of test item development and its various applications. Focus on Validity. Validity the most important consideration in testing is stressed throughout and is based on the Standards for Educational and Psychological Testing currently under revision by AERA, APA and NCME. Illustrative Examples. The book presents various selected and constructed response formats and uses many examples to illustrate correct and incorrect ways of writing items. Strategies for training item writers and developing large numbers of items using algorithms and other item generating methods are also presented. Based on Theory and Research. A comprehensive review and synthesis of existing research runs throughout the book and complements the expertise of its authors.

SAGE Handbook of Research on Classroom Assessment James H. McMillan, 2013. The Sage Handbook of Research on Classroom Assessment provides scholars, professors, graduate students and other researchers and policy makers in the organizations, agencies, testing companies and school districts with a comprehensive source of research on all aspects of K-12 classroom assessment. The handbook emphasizes theory, conceptual frameworks and all varieties of research: quantitative, qualitative, mixed methods to provide an in-depth understanding of the knowledge base in each area of classroom assessment and how to conduct inquiry in the area. It presents classroom assessment research to convey in depth the state of knowledge and understanding that is represented by the research with particular emphasis on how classroom assessment practices affect student achievement and teacher behavior. Editor James H. McMillan and five Associate Editors bring the best thinking and analysis from leading classroom assessment researchers on the nature of the research making significant contributions to this prominent and hotly debated topic in education.

Profiles of American Colleges -- 2008 Barron's Educational Series, 2008-07-01. Up-to-date facts and figures on enrollments, tuition and fees, academic programs, campus environment, available financial aid and much more make the 28th edition of Profiles of American Colleges America's most authoritative data source for college-bound high school students, their parents and high school guidance counselors. More than 1,650 accredited four-year colleges are profiled. An interactive CD-ROM enclosed with the directory guides students to specific schools when they enter details describing their personal academic plans and aptitudes. In addition to the above-cited information, each college profile gives details on Admission requirements, Library and computer facilities, Admissions procedures for freshmen, Campus safety and security, Thumbnail descriptions of faculty, Requirements for a degree, Athletic facilities, Extracurricular activities, E-mail addresses, College fax numbers and web sites, Admissions Contacts and much more. Schools are rated according to Barron's well-known competitiveness scale from Noncompetitive to Most Competitive. Unlike some other publications, Barron's refrains from the unreliable practice of ranking colleges on a first-through-last basis. The book's tinted pages section presents a quick reference Index of College Majors that lists all available major study programs at each school. Also profiled are many excellent colleges in Canada and several other

countries as well as brief profiles of religious colleges and American colleges based in foreign countries **The Publishers Weekly** ,2008

Constitutive Activity in Receptors and Other Proteins, Part B ,2010-11-24 This volume of *Methods in Enzymology* covers the current methodology for the detection and assessment of constitutively active proteins The chapters written by expert authors who are leaders in the field provide hints and tricks not available in primary research publications It is extensively referenced with useful figures and tables throughout the volume Expert authors who are leaders in the field Extensively referenced and useful figures and tables Provides hints and tricks to facilitate reproduction of methods *Guide to College Visits* Princeton Review,2007-04-10 A practical handbook covers everything needed to plan a successful visit to campuses across the country with information on nearby accommodations for every budget the best and worst times to visit itineraries and campus tours

Johnstone Triangle Norman Reid,2021-01-18 Chemistry is often seen as a difficult subject to understand This book focusses on the triangle model that Alex H Johnstone developed in the early 1980s Originally conceived in the context of making chemistry more accessible to a wider range of learners the model has been applied in almost every area of education in chemistry at all stages of learning In looking at why chemistry is difficult there are two central questions Firstly does the problem relate to the nature of chemistry and secondly does it relate to the way humans gain understanding Both were found to be important and the answers to the two question were found to be connected The triangle model arose from sustained research into human learning The central finding from research is the critical role of working memory and the model rationalises so much evidence from chemistry education research as well as the repeated experiences of teachers of chemistry at all levels In order to understand chemistry it is essential to develop sound mental models of molecular reality It generates major implications for the way a chemistry curriculum should be constructed and the processes of teaching and learning in chemistry when the goal is focussed on understanding the key ideas Some of these implications are developed and pointers offered to more successful ways forward The power of the Johnstone Triangle lies in the way it offers clear directions for all involved in chemistry education It is hoped that this book will prove helpful to all involved in sharing the exciting story of the way humans have come to understand the molecular world one of the great examples of great human endeavour

The New Visual Neurosciences John S. Werner,Leo M. Chalupa,2013-10-25 A comprehensive review of contemporary research in the vision sciences reflecting the rapid advances of recent years Visual science is the model system for neuroscience its findings relevant to all other areas This essential reference to contemporary visual neuroscience covers the extraordinary range of the field today from molecules and cell assemblies to systems and therapies It provides a state of the art companion to the earlier book *The Visual Neurosciences* MIT Press 2003 This volume covers the dramatic advances made in the last decade offering new topics new authors and new chapters *The New Visual Neurosciences* assembles groundbreaking research written by international authorities Many of the 112 chapters treat seminal topics not included in the earlier book These new topics include retinal feature detection cortical connectomics new approaches to mid level vision

and spatiotemporal perception the latest understanding of how multimodal integration contributes to visual perception new theoretical work on the role of neural oscillations in information processing and new molecular and genetic techniques for understanding visual system development An entirely new section covers invertebrate vision reflecting the importance of this research in understanding fundamental principles of visual processing Another new section treats translational visual neuroscience covering recent progress in novel treatment modalities for optic nerve disorders macular degeneration and retinal cell replacement The New Visual Neurosciences is an indispensable reference for students teachers researchers clinicians and anyone interested in contemporary neuroscience Associate Editors Marie Burns Joy Geng Mark Goldman James Handa Andrew Ishida George R Mangun Kimberley McAllister Bruno Olshausen Gregg Recanzone Mandyam Srinivasan W Martin Usrey Michael Webster David Whitney Sections Retinal Mechanisms and Processes Organization of Visual Pathways Subcortical Processing Processing in Primary Visual Cortex Brightness and Color Pattern Surface and Shape Objects and Scenes Time Motion and Depth Eye Movements Cortical Mechanisms of Attention Cognition and Multimodal Integration Invertebrate Vision Theoretical Perspectives Molecular and Developmental Processes Translational Visual Neuroscience

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