

66. which radiation are of very high frequency
a. I.R. b. X-ray c. Visible rays d. U.V
67. Poise is equal to.....
a. $1000\text{gcm}^{-1}\text{sec}^{-1}$ b. $100\text{Kgm}^{-1}\text{sec}^{-1}$ c. $10\text{Kgm}^{-1}\text{sec}^{-1}$ d. $0.1\text{Kgm}^{-1}\text{sec}^{-1}$
68. Isomorphous substance have same
a. B.P b. M.P c. atomic point d. lattice point
69. In a process 200J of heat is absorbed by a system while 200J work done by system. The change in internal energy
a. 200J b. 50J c. 100J d. 300J
70. Molecular orbital of lower energy than atomic orbital is called.....
a. Bonding orbital b. antibonding orbital c. Atomic orbital d. Non-bonding orbital
71. A sample of gas occupies 50dm³ at S.T.P assuming constant pressure what volume will the gas occupy if temp is double.....
a. 25dm³ b. 50dm³ c. 200dm³ d. 2500dm³
72. Example of trigonal system is... a. BaSO₄ b. AgNO₃ c. ZnSO₄ d. SiO₂ e. SnO₂
73. Four container of equal volume are filled as follows
I. 2gm of H₂ at 0°C II. 1g of H₂ at 27°C
III. 24gm of O₂ at 0°C IV. 36gm O₂. Which container is at greatest pressure.?
a. I b. II c. III d. IV
74. All of the following are non polar bond except. a. O₂ b. O₃ c. N₂ d. HF e. H₂
75. water has maximum density at temperature of... ? a. -4°C b. 4°C c. -20°C d. 100°C e. 30°C
76. Phenol phthalein is an indicator used for
a. strong acid & weak base b. strong base & weak acid
c. Strong acid & strong base d. weak acid & weak base.
77. which of the following will be the most electrically conductive....
a. sugar dissolved in water b. salt water c. salt dissolved in an organic compound
d. an oil & water mixture
78. which of the following is an element
a. tap water b. graphite c. sea water d. Brass e. Sulphuric acid.
79. To what volume must 10.0ml of 5.00M HCl be diluted to make a 0.5M HCl solution.....
a. 1 ml b. 50ml c. 100ml d. 500ml e. 5000ml
80. The property being able to add enthalpies is based on the
a. law of conservation of heat b. first law of thermodynamics c. law of constant composition d. E=mc²
e. Hess's Law
81. If ΔH of a reaction is - 500Kcal/mol it indicates the reaction...
a. Endothermic b. exothermic c. unstable d. in need of catalyst e. N.O.T
82. Which of the following is most likely to increase the rate of a reaction
a. decreasing the temperature b. reducing the surface area of reactant
c. decreasing the concentration of reactant in the reaction vessel d. reducing activation energy
e. +ve catalyst.

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