

Chapter 10 Review: Chemical Quantities & Calculations

(aka – mole & volume calculations, % composition, empirical & molecular formulas)

For the test, be sure that you are familiar, and able to complete, the problems below as well as the problems found on the following worksheets: "Molecular Weight and Mole Calculations," "Percent Comp. & Chemical Formula Calculations" and "Chapter 10 Sample Problems." Check the class website for answers to the review worksheet.

1. The air around you is made up of 20.95% oxygen. How many oxygen molecules are in 75.0 grams of air? The density of air is 1.19 g/L. *assume STP.*
2. How much helium (*grams*) would be needed to inflate a balloon to a volume of 5.50L? *assume STP.*
3. The molecular formula of an antibacterial drug is $C_{17}H_{18}FN_2O_5$. How many fluorine atoms are in 150mg of this drug?
4. A molecule is made up of 40.0% carbon, 6.67% hydrogen, and an unknown amount of oxygen. What is the molecular formula of the molecule if its molar mass is 90.0g/mole?
5. What is the empirical formula for a compound made up of 71.72% chlorine, 16.16% oxygen and 12.12% carbon? Calculate the molecular formula for the compound if it has a molar mass of 296.7g/mol.
6. How many molecules are in 8.42 Kilograms of C_2H_4 ?

10 Chemical Quantities Guided Practice Problems Answers 234326

Heather Hattori, Richard H. Langley



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Chemistry: 1,001 Practice Problems For Dummies (+ Free Online Practice) Heather Hattori, Richard H.

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Unveiling the Power of Verbal Art: An Emotional Sojourn through **10 Chemical Quantities Guided Practice Problems Answers 234326**

In some sort of inundated with monitors and the cacophony of fast conversation, the profound power and psychological resonance of verbal artistry often fade into obscurity, eclipsed by the constant assault of noise and distractions. Yet, set within the lyrical pages of **10 Chemical Quantities Guided Practice Problems Answers 234326**, a fascinating work of literary beauty that pulses with raw thoughts, lies an memorable trip waiting to be embarked upon. Composed by way of a virtuoso wordsmith, that exciting opus manuals readers on an emotional odyssey, lightly revealing the latent possible and profound influence embedded within the complicated web of language. Within the heart-wrenching expanse of this evocative analysis, we shall embark upon an introspective exploration of the book is main themes, dissect its captivating writing design, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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