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38 ChemActivity 6

Cyclohexane

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# ChemActivity 6

# Part A: Energy Diagrams

(What is the sign of  $\Delta H_{roution}$  for an exothermic reaction?)

#### Model 1A: The Slot Machine

Imagine a gambler sitting at a slot machine. Over the course of an hour the gambler puts \$100 into the machine and the machine pays out \$90. We know a net change ( $\Delta$ ) of ten dollars occurred, but what is the sign of this change in money ( $\Delta$ \$). To determine if  $\Delta$ \$ = +10 or -10 we need to decide if we calculate  $\Delta$ \$ from the perspective of the gambler or from the perspective of the slot machine.



To translate this analogy into chemistry remember that:

- Money = energy.
- The slot machine = the reaction.
- The gambler = the experimenter (you).
- All changes (Δ) will be calculated from the perspective of the reaction (a.k.a. the slot machine).

#### Critical Thinking Questions

- Using the convention listed above for calculating changes (Δ), what is the sign of ΔS in the example in Model 1A?
- Based on this same convention, what sign must be attached to the \$100 that the gambler put into the slot machine in Model 1A? The sign of the \$90 that comes out is the opposite. Add the appropriate sign to each value in Model 1A.
- On a different day the gambler puts in a total of \$20 and the slot machine pays out a total of \$100. What is the sign and value of ΔS for this day?





# **Answers Chemactivity 6**

Michael P. Garoutte, Ashley B. Mahoney

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