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0620 CHEMISTRY

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Paper 62



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(b) (i) The student adds about 10 cm³ of solution C to solution A and followed by a few drops of aqueous sodium nitrate.

Observations ... No change.

..... (1)

(ii) To the second portion of solution C, the student adds about 10 cm³ of dilute nitric acid followed by a few drops of aqueous sodium nitrate.

Observations ... Silver precipitate.

..... (1)

(c) (i) To the first portion of solution C, the student adds aqueous of aqueous sodium hydroxide.

Observations ... No change.

..... (1)

(ii) The student adds the product from part (b) (ii) onto a piece of

litmus paper. The colour of

litmus paper ... Red litmus paper turns blue.

★	Test 1 ★	Test 2 ★	Test 3 ★
Liquid Solution	Red	Blue	Purple
Litmus Paper (Red)	Turns Red	Remains	No change
Litmus Paper (Blue)	Turns Red	Turns Blue	No change
Acidic Orange	Red	Yellow	Orange
Phenolphthalein	Colourless	Pink	Colourless
Thymolphthalein	Colourless	Blue	Colourless

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0620/62/24

Page 10 of 10

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K Morrison



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1. Understanding the eBook 0620 12 O N 11 Ms
 - The Rise of Digital Reading 0620 12 O N 11 Ms
 - Advantages of eBooks Over Traditional Books
2. Identifying 0620 12 O N 11 Ms
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 0620 12 O N 11 Ms
 - User-Friendly Interface
4. Exploring eBook Recommendations from 0620 12 O N 11 Ms
 - Personalized Recommendations
 - 0620 12 O N 11 Ms User Reviews and Ratings
 - 0620 12 O N 11 Ms and Bestseller Lists
5. Accessing 0620 12 O N 11 Ms Free and Paid eBooks

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 - 0620 12 O N 11 Ms eBook Subscription Services
 - 0620 12 O N 11 Ms Budget-Friendly Options
6. Navigating 0620 12 O N 11 Ms eBook Formats
- ePub, PDF, MOBI, and More
 - 0620 12 O N 11 Ms Compatibility with Devices
 - 0620 12 O N 11 Ms Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of 0620 12 O N 11 Ms
 - Highlighting and Note-Taking 0620 12 O N 11 Ms
 - Interactive Elements 0620 12 O N 11 Ms
8. Staying Engaged with 0620 12 O N 11 Ms
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 0620 12 O N 11 Ms
9. Balancing eBooks and Physical Books 0620 12 O N 11 Ms
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection 0620 12 O N 11 Ms
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 0620 12 O N 11 Ms
- Setting Reading Goals 0620 12 O N 11 Ms
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 0620 12 O N 11 Ms
- Fact-Checking eBook Content of 0620 12 O N 11 Ms
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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