5 Carbonyl chloride, COC5, is a colourless gas. It is made by the following reaction.

Figs.

(a) When the pressure on the equilibrium mixture is decreased, the position of equilibrium moves to left.

(ii) How does the concentration of each of the three chemicals change?

		6.00

(W) Explain why the position of equilibrium moves to left.

(b) Using the information given with the equation, is the forward reaction exothermic or endothermic? Give a reason for your choice.

97

(c) Carbonyl chloride reacts with water to form two acidic compounds. Suggest which acidic compounds are formed.

100			
			 -

(d). The structural formula of carbonyl chloride is given below.



Draw a diagram that shows the arrangement of the valency electrons in one molecule of this covalent compound.

Use x for an electron from a chlorine atom.

Use o for an electron from a carbon atom.

Use . For an electron from an oxygen atom.

[Total: 12]

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