

Page 2	Mark Scheme: Teachers' version IGCSE – May/June 2011	Syllabus 0620	Paper 31
1	(a) F or B diffusion / <u>fractional</u> distillation		[1]
	(b) A simple distillation		[1]
	(c) D chromatography		[1]
	(d) E filtration		[1]
	(e) C evaporation		[1]
	(f) B <u>fractional</u> distillation		[1]
2	(a) (i) photosynthesis or a photochemical reaction not an example, question requires a process not devices which convert light into electricity		[1]
	(ii) cell accept battery not generator		[1]
	(b) (i) correct formula cond following marks conditional on correct formula if covalent mark 1 only correct charges 6s and 2p around anion do NOT penalise for incorrect coding ignore electrons around potassium		[1] [1] [1]
	(ii) correct formula if ionic mark 1 only cond 2 bp and 2 nbp around selenium 1 bp and 3 nbp around both chlorine atoms		[1] [1]
	(iii) the ionic compound higher melting point / boiling point / less volatile conducts when molten or aqueous, covalent compound does not is soluble in water, covalent is not / ionic insoluble in organic solvents, covalent soluble in organic solvents harder any two note there has to be comparison between the ionic compound and the covalent compound not density		[2]

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