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- (b) Add a suitable scale to the y-axis in Fig. 2.1. Your scale should extend to 2°C above your highest temperature in Table 2.1.

Plot your results from Experiments 1 to 6 on the grid.

Draw two straight lines through your points, one through the first three points, and one through the last three points. Extend your straight lines so that they cross.

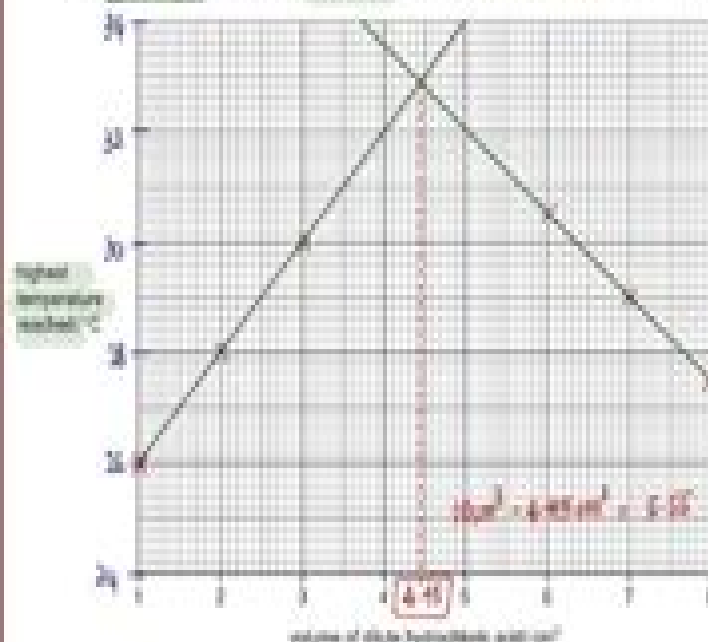


Fig. 2.1

(3)

- (c) The point on the graph where the two straight lines cross is where all of the aqueous sodium hydroxide reacts with all of the dilute hydrochloric acid to form a neutral solution.

- (i) Use your graph in Fig. 2.1 to deduce the volume of dilute hydrochloric acid and the volume of aqueous sodium hydroxide that react together to produce a neutral solution. Show your working on Fig. 2.1.

volume of dilute hydrochloric acid = 4.4 cm^3 ↓

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Table of Contents 0620 O N 14 62 Ms

1. Understanding the eBook 0620 O N 14 62 Ms
 - The Rise of Digital Reading 0620 O N 14 62 Ms
 - Advantages of eBooks Over Traditional Books
2. Identifying 0620 O N 14 62 Ms
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 0620 O N 14 62 Ms
 - User-Friendly Interface
4. Exploring eBook Recommendations from 0620 O N 14 62 Ms
 - Personalized Recommendations
 - 0620 O N 14 62 Ms User Reviews and Ratings

- 0620 O N 14 62 Ms and Bestseller Lists
- 5. Accessing 0620 O N 14 62 Ms Free and Paid eBooks
 - 0620 O N 14 62 Ms Public Domain eBooks
 - 0620 O N 14 62 Ms eBook Subscription Services
 - 0620 O N 14 62 Ms Budget-Friendly Options
- 6. Navigating 0620 O N 14 62 Ms eBook Formats
 - ePub, PDF, MOBI, and More
 - 0620 O N 14 62 Ms Compatibility with Devices
 - 0620 O N 14 62 Ms Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 0620 O N 14 62 Ms
 - Highlighting and Note-Taking 0620 O N 14 62 Ms
 - Interactive Elements 0620 O N 14 62 Ms
- 8. Staying Engaged with 0620 O N 14 62 Ms
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 0620 O N 14 62 Ms
- 9. Balancing eBooks and Physical Books 0620 O N 14 62 Ms
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 0620 O N 14 62 Ms
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 0620 O N 14 62 Ms
 - Setting Reading Goals 0620 O N 14 62 Ms
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 0620 O N 14 62 Ms
 - Fact-Checking eBook Content of 0620 O N 14 62 Ms
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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