

Name _____

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SECTION THE PERIODIC TABLE IS A MAP OF THE ELEMENTS.

5.3 Reinforcing Key Concepts

BIG IDEA A substance's atomic structure determines its physical and chemical properties.

KEY CONCEPT The periodic table is a map of the elements.

1. **The periodic table has distinct regions.** The periodic table is a map of the elements. Where an element is on the periodic table will give you an indication of its properties. What are the three main regions of the periodic table? What does *reactive* mean?

2. **Most elements are metals.** Most of the elements on the periodic table are metals. These metals are divided into groups according to their properties. Complete the chart below by describing the difference between the alkali, alkaline, and transition metals.

	Alkali Metals	Alkaline Metals	Transition Metals
Reactivity			
Examples			

3. **Nonmetals and metalloids have a wide range of properties.** Use a periodic table to identify each of the elements below as a nonmetal, halogen, noble gas, or metalloid. Some elements can have more than one answer.
- Silicon _____
 - Chlorine _____
 - Nitrogen _____
 - Argon _____
4. **Some atoms can change their identity.** Sometimes, an atom's nucleus may have too many or too few neutrons to be stable. When this occurs, the atom will produce particles and energy until it is again stable. If the number of protons in the nucleus changes, the identity of the atom changes. Atoms that change identity are said to be radioactive. Scientists measure radioactivity by the time it takes for one-half of a sample of atoms to change identity. For example, lead-214 has a half-life of 27 minutes. If you started with 500 grams of this isotope, how many grams would you have after 54 minutes?

1 1 Reinforcing Key Cell Test Concepts Science

**Catherine A. Sanderson, Karen R.
Huffman**



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Table of Contents 1 1 Reinforcing Key Cell Test Concepts Science

1. Understanding the eBook 1 1 Reinforcing Key Cell Test Concepts Science
 - The Rise of Digital Reading 1 1 Reinforcing Key Cell Test Concepts Science
 - Advantages of eBooks Over Traditional Books
2. Identifying 1 1 Reinforcing Key Cell Test Concepts Science
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 1 1 Reinforcing Key Cell Test Concepts Science
 - User-Friendly Interface
4. Exploring eBook Recommendations from 1 1 Reinforcing Key Cell Test Concepts Science
 - Personalized Recommendations
 - 1 1 Reinforcing Key Cell Test Concepts Science User Reviews and Ratings

- 1 1 Reinforcing Key Cell Test Concepts Science and Bestseller Lists
- 5. Accessing 1 1 Reinforcing Key Cell Test Concepts Science Free and Paid eBooks
 - 1 1 Reinforcing Key Cell Test Concepts Science Public Domain eBooks
 - 1 1 Reinforcing Key Cell Test Concepts Science eBook Subscription Services
 - 1 1 Reinforcing Key Cell Test Concepts Science Budget-Friendly Options
- 6. Navigating 1 1 Reinforcing Key Cell Test Concepts Science eBook Formats
 - ePub, PDF, MOBI, and More
 - 1 1 Reinforcing Key Cell Test Concepts Science Compatibility with Devices
 - 1 1 Reinforcing Key Cell Test Concepts Science Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 1 1 Reinforcing Key Cell Test Concepts Science
 - Highlighting and Note-Taking 1 1 Reinforcing Key Cell Test Concepts Science
 - Interactive Elements 1 1 Reinforcing Key Cell Test Concepts Science
- 8. Staying Engaged with 1 1 Reinforcing Key Cell Test Concepts Science
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 1 1 Reinforcing Key Cell Test Concepts Science
- 9. Balancing eBooks and Physical Books 1 1 Reinforcing Key Cell Test Concepts Science
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 1 1 Reinforcing Key Cell Test Concepts Science
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 1 1 Reinforcing Key Cell Test Concepts Science
 - Setting Reading Goals 1 1 Reinforcing Key Cell Test Concepts Science
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 1 1 Reinforcing Key Cell Test Concepts Science
 - Fact-Checking eBook Content of 1 1 Reinforcing Key Cell Test Concepts Science
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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