

Now electronic configuration Co^{3+} :



Thus, outer orbital or high spin complex of $[\text{CoF}_6]^{3-}$:



(Six pairs of electrons from six fluoride ions.)

(b) Dibromide bis (ethane-1, 2-diamine) Cobalt (III) is the IUPAC name of $[\text{CoBr}_2(\text{en})_2]^+$ complex.

(c) The number of ions produced from the complex

$[\text{Co}(\text{NH}_3)_6]\text{Cl}_2$ in solution is three.



7. (a) Differentiate between the following:*

[1 × 3 = 3]

- (i) Adsorption and Absorption
- (ii) Lyophobic Sol and Lyophilic Sol
- (iii) Multimolecular Colloid and Macromolecular colloid

OR

(b) (I) Define the following terms:*

- (i) Zeta Potential
- (ii) Coagulation

(II) Why a negatively charged sol is obtained when AgNO_3 solution is added to KI solution?*

[3]

8. Define transition metals. Why Zn, Cd and Hg are not called transition metals? How is the variability in oxidation states of transition metals different from that of p-block elements?

[3]

Ans. Transition metals are the metals that have partially filled d-orbitals and having the ability to form stable cations.

Zn, Cd, Hg are not regarded as transition metals because they have completely filled d-subshell with outer electronic configuration $(n-1)d^{10}ns^2$.

The variability in oxidation states of transition metals is due to the incomplete filling of d-orbitals in such a way, that their oxidation states differ from each other by unity (i.e., Fe^{2+} and Fe^{3+}). While, in case of p-block elements, the oxidation state differ by two units (i.e., +3 and +5).

9. (a) What happens when:

[1 × 3 = 3]

(i) Propanone is treated with CH_3MgBr and then hydrolysed?

(ii) Ethanal is treated with excess ethanol and acid?

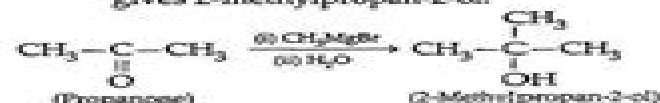
(iii) Methanal undergoes Cannizzaro reaction?

OR

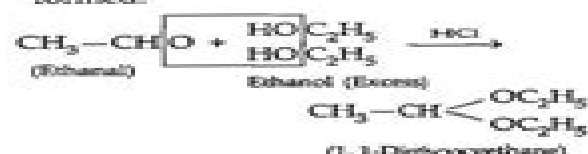
(b) Write the main product in the following reactions: [1 × 3 = 3]



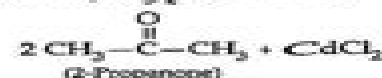
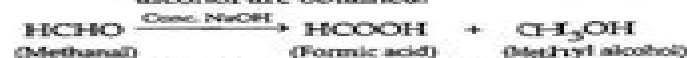
Ans. (a) (i) When propanone is treated with CH_3MgBr followed by hydrolysis it gives 2-methylpropan-2-ol.



(ii) When ethanal reacts with excess of ethanol in presence of trace amount of HCl then, 1, 1-Dithoxyethane will be formed.



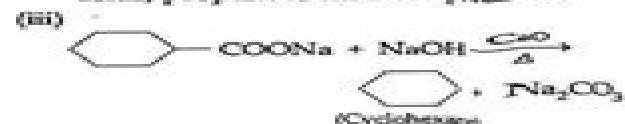
(iii) When methanal undergoes Cannizzaro reaction then, formic acid and methyl alcohol are obtained.



Thus, 2-Propanone is the main product.



Thus, propane is the main product.



Thus, cyclohexane is the main product.

10 Years 12 Board Question Paper Chemistry

K Morrison



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