

2. An iron bar rusts. The iron reacts with oxygen in the air to make rust.



Which chemical formula(s) represents the *reactants*?



Which chemical formula(s) represents the *products*?



How many of each atom is present in the *reactants*?

Fe - 4 ✓
O - 6 ✓

How many of each atom is present in the *products*?

Fe - 4 ✓
O - 6 ✓

Is this a balanced equation? Explain.

Yes! The reactants = the products



Analyzing Chemical Equations Answer Key

B Lingard



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Mastering NEET MCQs: Speed-solving Techniques for Medical Aspirants Dr. Ashokan Kannarath, 2024-01-31 This book dives deep into the strategies techniques and tips required to excel in the NEET MCQs It is designed to help you not only solve questions quickly but also enhance your understanding and retention of vital medical concepts It explains how to analyse NEET MCQs without getting overwhelmed efficiently identifying keywords understanding question patterns and eliminating wrong answer options It also exposes proven strategies to increase your answering speed without compromising accuracy Includes time management tips mental math tricks and effective test taking methods Chemical Engineering Thermodynamics AHUJA, PRADEEP, 2008-12 This book offers a full account of thermodynamic systems in chemical engineering It provides a solid understanding of the basic concepts of the laws of thermodynamics as well as their applications with a thorough discussion of phase and chemical reaction equilibria At the outset the text explains the various key terms of thermodynamics with suitable examples and then thoroughly deals with the virial and cubic equations of state by showing the P V T pressure molar volume and temperature relation of fluids It elaborates on the first and second laws of thermodynamics and their applications with the help of numerous engineering examples The text further discusses the concepts of exergy standard property changes of chemical reactions thermodynamic property relations and fugacity The book also includes detailed discussions on residual and excess properties of mixtures various activity coefficient models local composition models and group contribution methods In addition the text focuses on vapour liquid and other phase equilibrium calculations and analyzes chemical reaction equilibria and adiabatic reaction temperature for systems with complete and incomplete conversion of reactants Key Features Includes a large number of fully worked out examples to help students master the concepts discussed Provides well graded problems with answers at the end of each chapter to test and foster students conceptual understanding of the subject The total number of solved examples and end chapter exercises in the book are over 600 Contains chapter summaries that review the major concepts covered The book is primarily designed for the undergraduate students of chemical engineering and its related disciplines such as petroleum engineering and polymer engineering It can also be useful to professionals The Solution Manual containing the complete worked out solutions to chapter end exercises and problems is available for instructors Chemical Engineering Design and Analysis T. Michael Duncan, Jeffrey A. Reimer, 2019-01-24 The go to guide to learn the principles and practices of design and analysis in chemical engineering *Quantitative Chemical Analysis* Mr. Rohit Manglik, 2024-07-11 EduGorilla Publication is a trusted name in the education sector committed to empowering learners with high quality study materials and resources Specializing in competitive exams and academic support EduGorilla provides comprehensive and well structured content tailored to meet the needs of students across various streams and levels **(Free Sample) Inorganic Qualitative Analysis for JEE Advanced/ Main & Boards (Class 11 & 12)** Dr. Deepak Joshi, 2021-11-26 The book Inorganic Qualitative Analysis is a rare

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