

Page 4	Mark Scheme: Teachers' version IGCSE – October/November 2009	Syllabus 0620	Paper 32
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- (b) zinc is more reactive [1]
 it loses electrons and forms ions in preference to iron [1]
 zinc corrodes not iron [1]
NOT zinc rusts

OR zinc loses electrons and forms ions [1]
 the electrons move on to the iron [1]
 the iron cannot be oxidised **or** it cannot rust **or** it cannot lose electrons [1]
CREDIT correct Chemistry that includes the above ideas

- (c) (i) zinc atoms change into ions, (the zinc dissolves) [1]
 copper(II) ions change into atoms, (becomes plated with copper) [1]

- (ii) ions [1]
 electrons [1]

[Total: 10]

- 4 (a) diffusion [1]
 different M_r **or** ozone molecules heavier than oxygen molecules
or different densities **or** oxygen molecules move faster than ozone molecules [1]
NOT oxygen is lighter **or** ozone heavier

OR fractional distillation [1]
 they have different boiling points [1]

- (b) (i) from colourless (solution) [1]
 to brown (solution) [1]

- (ii) I^- loses electrons (it is oxidised) [1]

- (iii) they are accepted by ozone [1]
or ozone is an electron acceptor

- (c) (i) water
 carbon dioxide
 sulfur dioxide
 all three [2]
 any two [1]

- (ii) correct structural skeleton [1]
COND 4bp around both carbon atoms [1]
 2bp and 2bp around sulfur atom [1]

[Total: 11]

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Transvaals



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