

7-1 Review and Reinforcement

Ionic Bonding

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

negative
positive
crisscross

noble gas
electrons
anion

octet
ions
cation

electron configuration
polyatomic
empirical formula

1. An ionic bond is an attraction between oppositely charged ions.
2. Anions have a negative charge.
3. An atom becomes an ion by losing or gaining electrons.
4. The octet rule states that atoms tend to gain, lose, or share electrons in order to acquire a full set of valence electrons.
5. When sodium and chlorine form an ionic bond, both ions acquire the electron configuration of a(n) noble gas.
6. The atoms that make up polyatomic ions are bonded together by covalent bonds.
7. The crisscross method can be used to write the formula for an ionic compound.
8. The empirical formula of a compound denotes the ratio of ions in the compound.

If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line.

losing

true

true

polyatomic

seven

true (released)

9. Calcium becomes a monatomic cation by gaining two electrons.
10. A cation has a positive charge.
11. A binary ionic compound contains only one kind of cation and one kind of anion.
12. Monatomic ions consist of more than one atom.
13. The Lewis dot diagram for chlorine, a group 7A element, has six electrons.
14. A great deal of energy is produced when an ionic compound is formed.

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Kaplan & Sadock's Study Guide and Self-Examination Review in Psychiatry Benjamin J. Sadock, Virginia A. Sadock, Pedro Ruiz, 2012-03-28 Kaplan Sadock's Study Guide and Self Examination Review in Psychiatry is a comprehensive review of the specialty and perfect for stand alone review or as preparation for the PRITE in service ABPN Part I and recertification examinations The book contains more than 1 600 multiple choice questions and answers with explanatory discussions of correct and incorrect responses Chapters parallel the essential corresponding chapters in Kaplan Sadock's Synopsis of Psychiatry a staple of psychiatry education around the globe Terms and definitions are consistent with DSM IV TR and ICD 10

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Table of Contents 10 1 Review And Reinforcement Continued Answers

1. Understanding the eBook 10 1 Review And Reinforcement Continued Answers
 - The Rise of Digital Reading 10 1 Review And Reinforcement Continued Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying 10 1 Review And Reinforcement Continued Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 10 1 Review And Reinforcement Continued Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from 10 1 Review And Reinforcement Continued Answers
 - Personalized Recommendations
 - 10 1 Review And Reinforcement Continued Answers User Reviews and Ratings
 - 10 1 Review And Reinforcement Continued Answers and Bestseller Lists
5. Accessing 10 1 Review And Reinforcement Continued Answers Free and Paid eBooks
 - 10 1 Review And Reinforcement Continued Answers Public Domain eBooks
 - 10 1 Review And Reinforcement Continued Answers eBook Subscription Services
 - 10 1 Review And Reinforcement Continued Answers Budget-Friendly Options
6. Navigating 10 1 Review And Reinforcement Continued Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - 10 1 Review And Reinforcement Continued Answers Compatibility with Devices
 - 10 1 Review And Reinforcement Continued Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 10 1 Review And Reinforcement Continued Answers
 - Highlighting and Note-Taking 10 1 Review And Reinforcement Continued Answers
 - Interactive Elements 10 1 Review And Reinforcement Continued Answers
8. Staying Engaged with 10 1 Review And Reinforcement Continued Answers

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 10 1 Review And Reinforcement Continued Answers
9. Balancing eBooks and Physical Books 10 1 Review And Reinforcement Continued Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection 10 1 Review And Reinforcement Continued Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 10 1 Review And Reinforcement Continued Answers
- Setting Reading Goals 10 1 Review And Reinforcement Continued Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 10 1 Review And Reinforcement Continued Answers
- Fact-Checking eBook Content of 10 1 Review And Reinforcement Continued Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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