

1996 1996 1996 1996 1996 1996



# C/K Truck

## SERVICE MANUAL

BOOK 1 OF 2

Includes all except:  
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**GMT/96-CK-1**

# 1996 Chevy K1500 Service Manual

**D Kirk**



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web 1 molarity  $m = \frac{m}{M}$  mass molar mass  $\times 0.4500 \text{ L} = 0.825 \text{ g}$  141.9579 g/mol  $\times 0.0129 \text{ M} = 2$  molality  $0.825 \text{ g} / 141.9579 \text{ g/mol} = 0.00581158 \text{ mol} / 0.00581158 \text{ mol} = 0.4500 \text{ kg} = 0.0129 \text{ M}$  3 mole fraction  $n_2 = \frac{0.825 \text{ g} / 141.9579 \text{ g/mol}}{0.825 \text{ g} / 141.9579 \text{ g/mol} + 0.00581158 \text{ mol} \times 18.015 \text{ g/mol}} = 0.0129$

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m mix m 1 v 1 m 2 v 2 v 1 v 2 1 5 x 480 1 2 x 520 480 520 720 624 1000 1 344 m q5 with increase in temperature which of these changes a molality b fraction of solute present in water

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