

Specific Heat and Heat Capacity

Name _____ KEY _____ Period _____

1. Calculate the amount of heat needed to increase the temperature of 250 g of water from 20 °C to 46 °C.

$$\begin{aligned}q &= m \times C \times \Delta T \\q &= 250 \text{ g} \times 4.18 \text{ J/g}^\circ\text{C} \times 26^\circ\text{C} \\q &= 27,170 \text{ J or } 27 \text{ kJ}\end{aligned}$$

2. Calculate the specific heat capacity of copper given that 204.75 J of energy raises the temperature of 15 g of copper from 25 °C to 60. °C.

$$\begin{aligned}q &= m \times C \times \Delta T \\C &= q/m \times \Delta T \\C &= 204.75 \text{ J} / (15 \text{ g} \times 35^\circ\text{C}) \\C &= 0.39 \text{ J/g}^\circ\text{C}\end{aligned}$$

3. 216 J of energy is required to raise the temperature of aluminum from 15 °C to 35 °C. Calculate the mass of aluminum. (Specific heat capacity of aluminum is 0.90 J/g °C).

$$\begin{aligned}q &= m \times C \times \Delta T \\m &= q/C \times \Delta T \\m &= 216 \text{ J} / (0.90 \text{ J/g}^\circ\text{C} \times 20^\circ\text{C}) \\m &= 12 \text{ g}\end{aligned}$$

4. The initial temperature of 150 g of ethanol was 22 °C. What will be the final temperature of the ethanol if 3240 J was needed to raise the temperature of the ethanol? (Specific heat capacity of ethanol is 2.44 J/g °C).

$$\begin{aligned}q &= m \times C \times \Delta T \\\Delta T &= q/m \times C \\\Delta T &= 3240 \text{ J} / (150 \text{ g} \times 2.44 \text{ J/g}^\circ\text{C}) \\\Delta T &= 8.85^\circ\text{C} \\T_{\text{final}} &= 22^\circ\text{C} + 8.85^\circ\text{C} = 31^\circ\text{C}\end{aligned}$$

5. When 435 J of heat is added to 3.4 g of olive oil at 21 °C, the temperature increases to 85 °C. What is the specific heat of the olive oil?

$$\begin{aligned}q &= m \times C \times \Delta T \\C &= q/m \times \Delta T \\C &= 435 \text{ J} / (3.4 \text{ g} \times 64^\circ\text{C}) \\C &= 2.0 \text{ J/g}^\circ\text{C}\end{aligned}$$

6. How many kilojoules of heat are absorbed when 1.00 L of water is heated from 18 °C to 85 °C? (Hint: You first need to determine the mass of the water, then calculate q in the requested unit.)

1 Specific Heat Answers

BM King

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'BM King' text.

1 Specific Heat Answers:

1 Specific Heat Answers Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**1 Specific Heat Answers**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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