- 13. (4 pts) What is an ionic compound?
 - a. A compound that will dissolve in water
 - b. Two nonmetal atoms bonded together
 - c. Two metal atoms bonded together
 - d. A metal and a nonmetal atom bonded together
 - e. None of the above
- 14. (4 pts) What type of compound uses the naming system consisting of prefixes like mono-, di-, and tri-?
 - a. Ionic
 - b. Covalent
 - c. Both a and b
 - d. Neither a nor b
- 15. (4 pts) What type of compound uses a naming system that has -ide as the ending on the nonmetals?
 - a. Ionic
 - b. Covalent
 - c. Both a and b
 - d. Neither a nor b
- 16. (4 pts) How many total atoms are in this molecule: C₆H₁₂OH?
 - a. 3
 - b. 4
 - c. 19
 - d. 20
 - e. None of the above
- 17. (4 pts) You are told that a molecule contains a total of 17 atoms. If the formula is C₁H₁₀Cl₂, what must x be?
 - a. 1
 - b. 2
 - c. 3
 - d. 4
 - e. None of the above
- (4 pts) Which of the following compounds would conduct electricity when dissolved in water? C₃H₆ClF C₆H₅CH₅ CS₂ BaSO₄
 - a. C₃H₆CIF
 - b. C₆H₃CH₃
 - e. CSa
 - d. BaSO₄
 - e. CS2 and BaSO4

Answers For Chemistry Test Module 13

Georgios Tsaparlis

Answers For Chemistry Test Module 13:

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Problems and Problem Solving in Chemistry **Education** Georgios Tsaparlis, 2021 Problem solving is central to the teaching and learning of chemistry at secondary

tertiary and post tertiary levels of education opening to students and professional chemists alike a whole new world for analysing data looking for patterns and making deductions As an important higher order thinking skill problem solving also constitutes a major research field in science education Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner El-Hi Textbooks in Print ,1981 **Engineering Chemistry** Thakur Publication, 2021-06-03 Buy Solved Series of Engineering Chemistry E Book for B Tech I II Semester Students Common to All of APJ Abdul Kalam Technological University KTU Kerala Revise As and A2 - Chemistry Rob Ritchie, 2008-10 Revise AS A2 Chemistry gives complete study support throughout the two A Level years This Study Guide matches the curriculum content and provides in depth course coverage plus invaluable advice on how to get the best results in the exams Nta Cuet (Pg)-2024 Chemistry Comprehensive Exam Guide | Including Latest Solved Paper & Mock Test Team Prabhat, 2024-02-09 NTA CUET PG 2024 CHEMISTRY COMPREHENSIVE GUIDE We present the NTA CUET PG 2024 CHEMISTRY COMPREHENSIVE GUIDE The book suffices the need of the aspirants in terms of Latest CUET Solved Paper 2023 Latest Examination Scheme and Syllabus Concise yet In depth Chapters Readability of the Content Concise yet In depth Chapters Ample figures and diagrams Solved MCQs Mock Test with Every Module Moreover the book is supplemented with a Joint Admission Test for Masters JAM Mock Test CHEMISTRY The book is divided into 3 Parts consisting chapters in detail PART I Inorganic Chemistry Module I comprises Periodic Table Chemical Bonding and Shapes of Compounds Main Group Elements Transmission Elements Module II comprises Bioinorganic Chemsitry Instrumental Menthods of Analysis Analytical Chemistry PART II Organic Chemistry Module I comprises Basic Concepts of Organic Chemistry and strerochemistry Organice Reaction Mechanism amd Synthetic Application Module II comprises Qualitative Organic Analysis Natural Products Chemistry Aromatic and Heterocyclic Chemistry PART III Physical Chemistry Module I comprises Basic Mathematical Concepts Atomic and Molecular Structure Theory of Gases Solid State Chemical Thermodynamics Module II comprises Chemical and Phase Equilibria Electrochemistry Chemical Kinetics Adsorption Spectroscopy This book serves to be a suitable Study Guide for the aspirants with focus on Qualitative Preparation and Systematic understanding of the Syllabus and

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