

Surface Weather Map Showing Storm System in the Eastern United States



Figure 2. Simplified surface weather map showing a storm system in the eastern US.

Warm air advection occurs where the wind flows across isobars from higher pressure to lower (warmer) values. Thus, warm air is being transported towards a low-pressure by the horizontal winds. Based on this fact alone, one would expect precipitation at that location to be rising, cold air advection occurs where the wind flows across the isobars from lower (colder) values to higher (warmer) values. Thus, cold air is being transported by the wind to a warm location, producing falling temperatures.

Based on wind directions and the isobars pattern on the Figure 2 map, determine the type of air advection (warm or cold) that would be occurring at each station.

3. Warm air advection was occurring at Station _____.

- ☐ A
☐ B
☐ C
☐ D

4. Cold air advection was occurring at Station _____.

- ☐ B
☐ C
☐ D

5. Considering the air mass displacements of wind used to determine pressure associated with weather systems, what is the most likely weather expected at Station _____.

- ☐ warm
☐ cold

6. Most likely, where on the west end coastline of Lake Erie the expected to have _____ air advection.

- ☐ warm
☐ cold

7. Areas to the east of High A would be expected to have _____ air advection.

- ☐ warm
☐ cold

8. And while you choose here that only the weather based on the model of a high, areas to the east of High A would have _____ air advection.

- ☐ warm
☐ cold

Answer Key Weather Studies Investigation Manual

Peipei Pang



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 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \quad \Delta h = 242 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) \rightarrow \text{C}_2\text{H}_4(\text{g}) + \text{H}_2(\text{g}) \quad \Delta h = 137 \text{ kJ}$
 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_6(\text{g}) + 1 \text{C}_2\text{H}_6(\text{g}) \quad \Delta h = 283 \text{ kJ}$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 2 \text{H}_2\text{O}(\text{l}) \quad \Delta h = -1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6(\text{g}) + 7 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{l}) \quad \Delta h = -1560 \text{ kJ/mole}$
 $\text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l}) \quad \Delta h = -285 \text{ kJ/mole}$

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