

The Atoms Family Atomic Math Challenge

Name _____



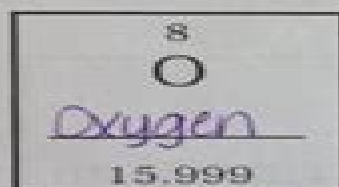
← Atomic Number
 ← Symbol
 ← Name
 ← Atomic Mass

Atomic number equals
the number of

protons or electrons

Atomic mass equals
the number of

protons + neutrons



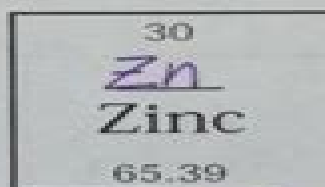
Atomic # = 8

Atomic Mass = 16

of Protons = 8

of Neutrons = 8

of Electrons = 8



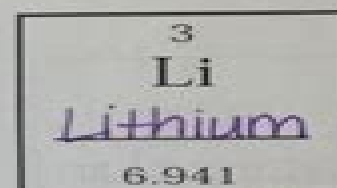
Atomic # = 30

Atomic Mass = 65

of Protons = 30

of Neutrons = 35

of Electrons = 30



Atomic # = 3

Atomic Mass = 7

of Protons = 3

of Neutrons = 4

of Electrons = 3



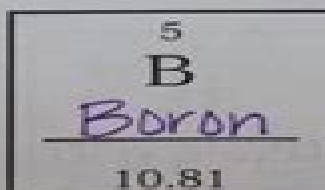
Atomic # = 14

Atomic Mass = 28

of Protons = 14

of Neutrons = 14

of Electrons = 14



Atomic # = 5

Atomic Mass = 11

of Protons = 5

of Neutrons = 6

of Electrons = 5



Atomic # = 35

Atomic Mass = 80

of Protons = 35

of Neutrons = 45

of Electrons = 35

Answer Key To The Atoms Family Atomic Math Challenge

RS Peters



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