

Types of Chemical Reaction Worksheet

Balance the reactions **1 to 6** and indicate which type of chemical reaction (synthesis, decomposition, single-displacement, double-displacement or combustion) is being represented:

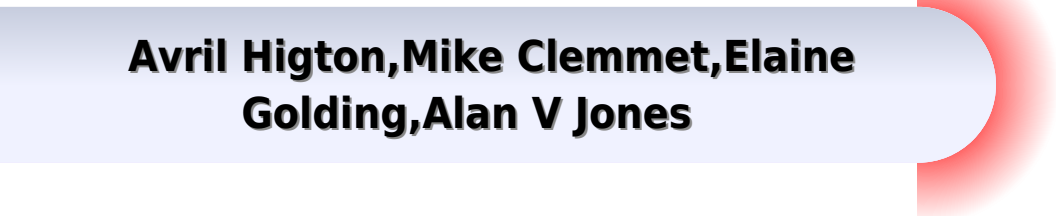
1. $2 \text{NaBr} + \text{Ca(OH)}_2 \rightarrow \text{CaBr}_2 + 2 \text{NaOH}$ Reaction Type : single displacement
2. $2 \text{NH}_3 + \text{H}_2\text{SO}_4 \rightarrow 3 (\text{NH}_4)_2\text{SO}_4$ Reaction Type : synthesis
3. $4 \text{C}_3\text{H}_8\text{O} + 27 \text{O}_2 \rightarrow 20 \text{CO}_2 + 18 \text{H}_2\text{O}$ Reaction Type : combustion
4. $3 \text{Pb} + 2 \text{H}_3\text{PO}_4 \rightarrow 3 \text{H}_2 + \text{Pb}_3(\text{PO}_4)_2$ Reaction Type : single displacement
5. $\text{Li}_3\text{N} + 3 \text{NH}_4\text{NO}_3 \rightarrow 3 \text{LiNO}_3 + (\text{NH}_4)_3\text{N}$ Reaction Type : double displacement
6. $\text{HBr} + 3 \text{Al(OH)}_3 \rightarrow \text{H}_2\text{O} + 3 \text{AlBr}_3$ Reaction Type : double displacement

Indicate which type of chemical reaction (synthesis, decomposition, single-displacement, double-displacement or combustion) is being represented in 7 to 20.

7. $\text{Na}_3\text{PO}_4 + 3 \text{KOH} \rightarrow 3 \text{NaOH} + \text{K}_3\text{PO}_4$ Reaction Type **double displacement**
8. $\text{MgCl}_2 + \text{Li}_2\text{CO}_3 \rightarrow \text{MgCO}_3 + 2 \text{LiCl}$ Reaction Type **double displacement**
9. $\text{C}_8\text{H}_{12} + 9 \text{O}_2 \rightarrow 6 \text{CO}_2 + 6 \text{H}_2\text{O}$ Reaction Type **combustion**
10. $\text{Pb} + \text{FeSO}_4 \rightarrow \text{PbSO}_4 + \text{Fe}$ Reaction Type **single displacement**
11. $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$ Reaction Type **decomposition**
12. $\text{P}_4 + 3 \text{O}_2 \rightarrow 2 \text{P}_2\text{O}_3$ Reaction Type **synthesis**
13. $2 \text{RbNO}_3 + \text{BeF}_2 \rightarrow \text{Be(NO}_3)_2 + 2 \text{RbF}$ Reaction Type **double displacement**
14. $2 \text{AgNO}_3 + \text{Cu} \rightarrow \text{Cu(NO}_3)_2 + 2 \text{Ag}$ Reaction Type **single displacement**
15. $\text{C}_3\text{H}_8\text{O} + 4 \text{O}_2 \rightarrow 3 \text{CO}_2 + 3 \text{H}_2\text{O}$ Reaction Type **combustion**
16. $2 \text{C}_3\text{H}_6 + \text{Fe} \rightarrow \text{Fe(C}_3\text{H}_5)_2$ Reaction Type **synthesis**
17. $\text{SeCl}_6 + \text{O}_2 \rightarrow \text{SeO}_2 + 3 \text{Cl}_2$ Reaction Type **single displacement**
18. $2 \text{MgI}_2 + \text{Mn(SO}_4)_2 \rightarrow 2 \text{MgSO}_4 + \text{MnI}_4$ Reaction Type **double displacement**
19. $\text{O}_3 \rightarrow \text{O} + \text{O}_2$ Reaction Type **decomposition**
20. $2 \text{NO}_2 \rightarrow 2 \text{O}_2 + \text{N}_2$ Reaction Type **decomposition**

Answers For Chemical Reactions

**Avril Higton, Mike Clemmet, Elaine
Golding, Alan V Jones**



Answers For Chemical Reactions:

Chemical Reactions, Materials and Particles Gill Murphy, 2006-03 These are a teaching resource designed for KS3 students whose literacy skills are considerably lower than their age. These books have an older format to counteract the simple text and cover various topics. They include activities, visuals, and assessment sheets as well as teacher pages.

Examination Questions and Answers in Basic Anatomy and Physiology Martin Caon, 2020-08-03 This third edition provides 2900 multiple choice questions on human anatomy and physiology and some biophysical science separated into 20 chapters and 68 categories. In addition, there are 64 essay topics. The answer to each question is accompanied by an explanation. Each chapter has an introduction to set the scene for the questions to come. However, not all possible information is provided within these introductions, so an Anatomy and Physiology textbook is an indispensable aid to understanding the answers. The textbook offers a more holistic approach to the subjects of anatomy and physiology by also including biomechanics, biophysics, and biochemistry. The questions have been used in end-of-semester examinations for undergraduate anatomy and physiology courses and as such reflect the focus of these particular courses and are pitched at this level to challenge students that are beginning their training in anatomy and physiology. The question and answer combinations are intended for use by teachers to select questions for their next examinations and by students when studying for an upcoming test. Students enrolled in the courses for which these questions were written include nursing, midwifery, paramedic, physiotherapy, occupational therapy, nutrition, and dietetics, health sciences, exercise science, and students taking an anatomy and physiology course as an elective.

Regents Exams and Answers: Chemistry--Physical Setting Revised Edition Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron's Regents Exams and Answers: Chemistry provides essential practice for students taking the Chemistry Regents, including actual recently administered exams and thorough answer explanations for all questions. This book features eight actual administered Regents Chemistry exams so students can get familiar with the test. Thorough explanations for all answers. Self-analysis charts to help identify strengths and weaknesses. Test-taking techniques and strategies. A detailed outline of all major topics tested on this exam. A glossary of important terms to know for test day.

Chemistry Equations And Answers (Speedy Study Guides) Speedy Publishing, 2015-04-24 Chemistry is a difficult subject to fully comprehend with its equations and scientific laws. Trying to digest an entire book in one semester is a tough job, but with the help of study guides like these, you can absorb information in chemistry much more effectively. This guide covers chemical equations, including examples, potential problems, and solutions.

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