

Use your notebook to complete all of the following review questions.

Please note that your completed notebook, corrected quizzes and unit review sheets are the **BEST** study tools. These questions will only help to reinforce some of the exam topics.

1. Draw a line continuum to show the metric prefixes in order from largest (on the left) to smallest (on the right).



2. Complete the following metric conversions:

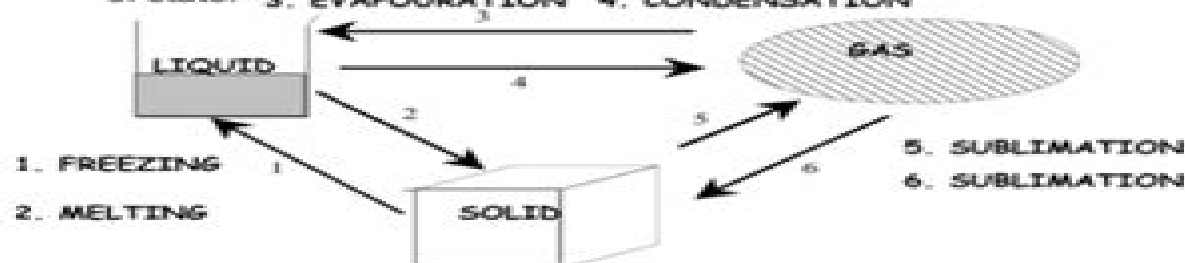
- a) 75 g = 0.075 kg                      e) 4.2 kL = 420 000 cL  
 b) 4530 cm = 0.0453 km                  f) 83.2 cm = 832 mm  
 c) 0.8 A = 800 mA                      g) 975 mg = 0.975 g  
 d) 3200 mA = 3.2 A                      h) 975 mg = 0.000975 kg

3. Complete the following table using your knowledge of the Particle Theory of Matter:

| STATE of MATTER | FORCES OF ATTRACTION | SPACING OF MOLECULES  | MOVEMENT OF MOLECULES |
|-----------------|----------------------|-----------------------|-----------------------|
| solid           | <b>strongest</b>     | <b>close together</b> | <b>least</b>          |
| liquid          | <b>weaker</b>        | <b>farther apart</b>  | <b>more</b>           |
| gas             | <b>weakest</b>       | <b>farthest apart</b> | <b>most</b>           |

4. Changes of state are examples of physical changes.

- a) Complete the following diagram with labels and arrows to show all of the SIX changes of state. 3. **EVAPOURATION** 4. **CONDENSATION**



5. Name FIVE ways in which you know a chemical change has probably taken place.

colour change                      light given off  
bubbles form                      heat produced  
precipitate forms                      \_\_\_\_\_

# Answers For Science Grade 9 Final Exam 2014

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### **Table of Contents Answers For Science Grade 9 Final Exam 2014**

1. Understanding the eBook Answers For Science Grade 9 Final Exam 2014
  - The Rise of Digital Reading Answers For Science Grade 9 Final Exam 2014
  - Advantages of eBooks Over Traditional Books
2. Identifying Answers For Science Grade 9 Final Exam 2014
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Answers For Science Grade 9 Final Exam 2014
  - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For Science Grade 9 Final Exam 2014
  - Personalized Recommendations
  - Answers For Science Grade 9 Final Exam 2014 User Reviews and Ratings
  - Answers For Science Grade 9 Final Exam 2014 and Bestseller Lists

5. Accessing Answers For Science Grade 9 Final Exam 2014 Free and Paid eBooks
  - Answers For Science Grade 9 Final Exam 2014 Public Domain eBooks
  - Answers For Science Grade 9 Final Exam 2014 eBook Subscription Services
  - Answers For Science Grade 9 Final Exam 2014 Budget-Friendly Options
6. Navigating Answers For Science Grade 9 Final Exam 2014 eBook Formats
  - ePub, PDF, MOBI, and More
  - Answers For Science Grade 9 Final Exam 2014 Compatibility with Devices
  - Answers For Science Grade 9 Final Exam 2014 Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Answers For Science Grade 9 Final Exam 2014
  - Highlighting and Note-Taking Answers For Science Grade 9 Final Exam 2014
  - Interactive Elements Answers For Science Grade 9 Final Exam 2014
8. Staying Engaged with Answers For Science Grade 9 Final Exam 2014
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Answers For Science Grade 9 Final Exam 2014
9. Balancing eBooks and Physical Books Answers For Science Grade 9 Final Exam 2014
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Answers For Science Grade 9 Final Exam 2014
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Answers For Science Grade 9 Final Exam 2014
  - Setting Reading Goals Answers For Science Grade 9 Final Exam 2014
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answers For Science Grade 9 Final Exam 2014
  - Fact-Checking eBook Content of Answers For Science Grade 9 Final Exam 2014
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning



- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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