

Chemistry - Review Chapters (415) 6

Directions: Answer each of the following questions. You need not use complete sentences.

1. By what property did Mendeleev arrange the elements?

ATOMIC MASS

2. By what property did Moseley suggest that the periodic table be arranged?

ATOMIC NUMBER

3. What is a period? How many are there in the periodic table?

A HORIZONTAL ROW IN THE PERIODIC TABLE; 7

4. What is a group (also called a family)? How many are there in the periodic table?

A VERTICAL COLUMN IN THE PERIODIC TABLE; 18

5. State the number of valence electrons in an atom of:

a. sulfur 6	b. calcium 2	c. chlorine 7	d. arsenic 5
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6. Give the names and chemical symbols for the elements that correspond to these atomic numbers:

a. 10 Ne, NEON	b. 18 Ar, ARGON	c. 36 Kr, KRYPTON	d. 90 Th, THORIUM
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7. List, by number, both the period and group of each of these elements.

	<u>Symbol</u>	<u>Period</u>	<u>Group</u>
a. beryllium	Be	2	2
b. iron	Fe	4	8
c. lead	Pb	6	14

8. Which of the following pairs of elements belong to the same period?

a. Na and Cl	b. Na and Li	c. Na and Cu	d. Na and Ne
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9. Which of the following pairs of elements belong to the same group?

a. H and He	b. Li and Be	c. C and Pb	d. Ga and Ge
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10. How does an element's period number relate to the number of the energy level of its valence electrons?

PERIOD NUMBER = ENERGY LEVEL OF VALENCE ELECTRONS

11. What are the transition elements?

GROUPS 3-12

12. In what type of orbitals are the actinide and lanthanide electrons found?

f ORBITALS

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Table of Contents Answers Prentice Hall Chemistry Ch6 Chemical Names Amd Formulas

1. Understanding the eBook Answers Prentice Hall Chemistry Ch6 Chemical Names Amd Formulas
 - The Rise of Digital Reading Answers Prentice Hall Chemistry Ch6 Chemical Names Amd Formulas
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers Prentice Hall Chemistry Ch6 Chemical Names Amd Formulas
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers Prentice Hall Chemistry Ch6 Chemical Names Amd Formulas
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers Prentice Hall Chemistry Ch6 Chemical Names Amd Formulas
 - Personalized Recommendations
 - Answers Prentice Hall Chemistry Ch6 Chemical Names Amd Formulas User Reviews and Ratings
 - Answers Prentice Hall Chemistry Ch6 Chemical Names Amd Formulas and Bestseller Lists
5. Accessing Answers Prentice Hall Chemistry Ch6 Chemical Names Amd Formulas Free and Paid eBooks
 - Answers Prentice Hall Chemistry Ch6 Chemical Names Amd Formulas Public Domain eBooks
 - Answers Prentice Hall Chemistry Ch6 Chemical Names Amd Formulas eBook Subscription Services
 - Answers Prentice Hall Chemistry Ch6 Chemical Names Amd Formulas Budget-Friendly Options
6. Navigating Answers Prentice Hall Chemistry Ch6 Chemical Names Amd Formulas eBook Formats

- ePub, PDF, MOBI, and More
 - Answers Prentice Hall Chemistry Ch6 Chemical Names And Formulas Compatibility with Devices
 - Answers Prentice Hall Chemistry Ch6 Chemical Names And Formulas Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers Prentice Hall Chemistry Ch6 Chemical Names And Formulas
 - Highlighting and Note-Taking Answers Prentice Hall Chemistry Ch6 Chemical Names And Formulas
 - Interactive Elements Answers Prentice Hall Chemistry Ch6 Chemical Names And Formulas
 8. Staying Engaged with Answers Prentice Hall Chemistry Ch6 Chemical Names And Formulas
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers Prentice Hall Chemistry Ch6 Chemical Names And Formulas
 9. Balancing eBooks and Physical Books Answers Prentice Hall Chemistry Ch6 Chemical Names And Formulas
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers Prentice Hall Chemistry Ch6 Chemical Names And Formulas
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Answers Prentice Hall Chemistry Ch6 Chemical Names And Formulas
 - Setting Reading Goals Answers Prentice Hall Chemistry Ch6 Chemical Names And Formulas
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Answers Prentice Hall Chemistry Ch6 Chemical Names And Formulas
 - Fact-Checking eBook Content of Answers Prentice Hall Chemistry Ch6 Chemical Names And Formulas
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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