In the reaction : 2Al₂O₃₍₁₎ → 4Al₍₈₎ + 3O_{2(g)}, the mole ratios are:

Al₂O₃ to Al _____ Al₂O₃ to O₂ _____ Al to O₂ ____ O₂ to Al _____

- 2. In the reaction: 2H₂ + O₂ → 2H₂O, the mole ratio of oxygen to water is: _____
- In the reaction: Ca + Cl₂ → CaCl₂, the mole ratio of chlorine to calcium is: ______
- In the reaction: Zn + H₂SO₄ → ZnSO₄ + H₂, the mole ratio of zinc to sulfuric acid is:
- The equation for the Haber process is represented by:

 $N_{2(g)} + 3H_{2(g)} \rightarrow 2NH_{3(g)}$

To convert 9.0 moles of hydrogen to ammonia would require how many moles of nitrogen?

 In the equation: 2KClO_{3(g)} → 2KCl_(s) + 3O_{2(g)}, how many moles of oxygen are produced when 3.00 moles of KClO₃ decompose completely?

 For the reaction: C₍₈₎ + 2H₂₍₉₎ → CH₄₍₉₎, how many moles of H₂ are required to produce 10.0 moles of CH₄?

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James E. Brady, John R. Holum

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