

POGIL: Periodic Trends in Atomic Properties

Information

Properties such as the size of an atom (atomic radius), the energy required to remove an electron from an atom (ionization energy), and the energy associated with the addition of an electron to a gaseous atom (electron affinity) can be understood in terms of the electron configuration of the atom and a competition between *electron-nucleus attraction* and *electron-electron repulsion*.

Preliminary Activity - Complete the orbital diagrams for the ground state configuration.

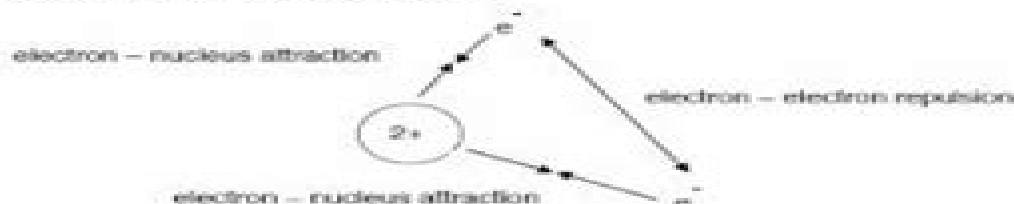
	1s	2s	2p		1s	2s	2p
H				C			
He				N			
Li				O			
Be				F			
B				Ne			

Model 1: Electron Ionization & Capture and Electrostatic Interactions

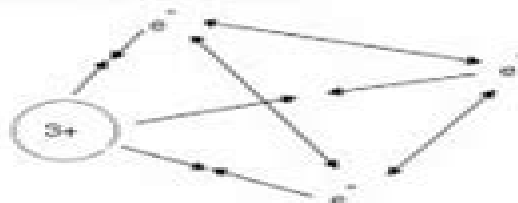
Reaction Equation for Electron Ionization: $X \rightarrow X^+ + e^-$ $\Delta E = 1^{\text{st}}$ ionization energy

Reaction Equation for Electron Capture: $X + e^- \rightarrow X^-$ $\Delta E = \text{electron affinity}$

Dominant inter-particle forces in helium:



Dominant inter-particle forces in lithium:



Answer Periodic Trends And Atomic Structure Pogil

Ainslie H. Drummond



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Table of Contents Answer Periodic Trends And Atomic Structure Pogil

1. Understanding the eBook Answer Periodic Trends And Atomic Structure Pogil
 - The Rise of Digital Reading Answer Periodic Trends And Atomic Structure Pogil
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Periodic Trends And Atomic Structure Pogil
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Periodic Trends And Atomic Structure Pogil
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Periodic Trends And Atomic Structure Pogil
 - Personalized Recommendations
 - Answer Periodic Trends And Atomic Structure Pogil User Reviews and Ratings
 - Answer Periodic Trends And Atomic Structure Pogil and Bestseller Lists

5. Accessing Answer Periodic Trends And Atomic Structure Pogil Free and Paid eBooks
 - Answer Periodic Trends And Atomic Structure Pogil Public Domain eBooks
 - Answer Periodic Trends And Atomic Structure Pogil eBook Subscription Services
 - Answer Periodic Trends And Atomic Structure Pogil Budget-Friendly Options
6. Navigating Answer Periodic Trends And Atomic Structure Pogil eBook Formats
 - ePub, PDF, MOBI, and More
 - Answer Periodic Trends And Atomic Structure Pogil Compatibility with Devices
 - Answer Periodic Trends And Atomic Structure Pogil Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Periodic Trends And Atomic Structure Pogil
 - Highlighting and Note-Taking Answer Periodic Trends And Atomic Structure Pogil
 - Interactive Elements Answer Periodic Trends And Atomic Structure Pogil
8. Staying Engaged with Answer Periodic Trends And Atomic Structure Pogil
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Periodic Trends And Atomic Structure Pogil
9. Balancing eBooks and Physical Books Answer Periodic Trends And Atomic Structure Pogil
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Periodic Trends And Atomic Structure Pogil
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answer Periodic Trends And Atomic Structure Pogil
 - Setting Reading Goals Answer Periodic Trends And Atomic Structure Pogil
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answer Periodic Trends And Atomic Structure Pogil
 - Fact-Checking eBook Content of Answer Periodic Trends And Atomic Structure Pogil
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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