
AQA BIOLOGY EXAM STYLE QUESTIONS WITH CORRECT DETAILED ANSWERS 2023

Explain the structure of phospholipids that makes them able to form the bilayer of cell membranes - Phospholipid heads are hydrophilic and the tail is hydrophobic.

They form a double layer with the hydrophilic heads facing the water and the hydrophobic tail facing inwards.
The center of the membrane bilayer is hydrophobic so water soluble substances cant easily pass through.

Describe the structure of a cellulose molecule and explain how cellulose is adapted for its function in cells. - 1. made from β -glucose;
2. joined by condensation / removing molecule of water / glycosidic bond;
3. 1 : 4 link specified or described;
4. "flipping over" of alternate molecules;
5. hydrogen bonds linking chains / long straight chains;
6. cellulose makes cell walls strong / cellulose fibres are strong;
7. can resist turgor pressure / osmotic pressure / pulling forces;
8. bond difficult to break;
9. resists digestion / action of microorganisms / enzymes;

Why is the genetic code described as being universal? - (In all organisms / DNA,) the same triplet codes for the same amino acid;

Accept codon / same three bases / nucleotides

Mutation 1 leads to the production of a non-functional protein. when mutatio occurson the exons

Explain why.

_____ - (Mutation) changes triplets / codons after that point / causes frame shift;

Accept changes splicing site

Ignore changes in sequence of nucleotides / bases

2. Changes amino acid sequence (after this) / codes for different amino acids (after this);

Accept changes primary structure

Aqa Biology Textbook Exam Style Questions Answers

**Christine Lea, Pauline Lowrie, Siobhan
McGuigan**



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web determine the shape of simple molecules determine the polarity of molecules using net molecular dipoles molecules have shapes there is an abundance of experimental evidence to that effect from their physical properties to their chemical reactivity

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web name lab 9 electron and molecular geometries polarity report sheet part d applying vsepr theory for each of the lewis structures shown below predict the electron geometry molecular geometry and bond angle

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web molecule is said to be polar if the overall net effect of all the dipole moments in each bond do not cancel each other out to determine polarity of a molecule we first determine the molecular geometry and then draw an arrow pointing from the positive to negative side of each bond the overall effect of each dipole is then considered after

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web structure geometry and polarity of molecules pre lab assignment name section 1 for each chemical formula below identify the compound as ionic or covalent and explain how you made the decision a sci b baci neon 2 indicate the covalent

bonding pattern for each of the non metal element listed below

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web when is a molecule polar change the electronegativity of atoms in a molecule to see how it affects polarity see how the molecule behaves in an electric field change the bond angle to see how shape affects polarity

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web answer all of these molecules and ions contain polar bonds only ClF_5 XeCl_2 PCl_3 SeF_4 and XePH_2 have dipole moments

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web in more complex molecules with polar covalent bonds the three dimensional geometry and the compound's symmetry determine whether there is a net dipole moment the dipole moment of a molecule is the vector sum of the dipole moments of
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web determine the lewis structure molecular geometry and polarity for molecules using lewis structures sketch molecular geometries on paper with wedges and dashes to represent their 3d shape explain how π bonding affects the ability of atoms to rotate about a bond relate molecular polarity to physical properties

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web jul 14 2012 this activity combines two simulations molecular shapes and molecular polarity to guide the students from drawing lewis structures to vsepr predicted geometries to predicting a molecule's polarity subject chemistry level undergrad intro type homework lab duration 90 minutes answers included no

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web oct 27 2022 two regions of electron density around a central atom in a molecule form a linear geometry three regions form a trigonal planar geometry four regions form a tetrahedral geometry five regions form a trigonal bipyramidal geometry and six regions form an octahedral geometry

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web sep 22 2021 molecular polarity molecular polarity results when the entire molecule not just a bond in the molecule ends up with an unequal distribution of electrons in general a molecule will be polar if it contains polar bonds that are distributed in a non symmetrical arrangement around the central atom a polar molecule is said to have a

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web sep 22 2021 vsepr theory the vsepr valence shell electron pair repulsion model is used to predict the geometry of molecules based on the number of effective electron pairs around a central atom

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web introduction molecular shape and polarity greatly influence the chemical and physical properties of compounds because

a molecule will always assume the geometry that gives it the lowest potential energy molecular shape can

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web draw lewis structure use vsepr to determine molecular geometry determine bond polarity based on electronegativity differences determine molecular polarity based on bond dipoles molecular geometry for the following molecules complete this

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web two regions of electron density around a central atom in a molecule form a linear geometry three regions form a trigonal planar geometry four regions form a tetrahedral geometry five regions form a trigonal bipyramidal geometry and six regions form an octahedral geometry

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web molecules composed of covalently bonded atoms may also be polar or nonpolar for the molecule to be polar it must of course have polar bonds but the key factor for determining the polarity of a molecule is its shape if the polar bonds dipoles are symmetrical around the central atom they offset each other and the resulting molecule is

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web linear geometry all angles 180 degrees three regions of electron density around a central atom in a molecule form a trigonal planar geometry all angles 120 degrees four regions of electron density around a central atom in a molecule form a tetrahedral geometry all angles 109.5 degrees

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web in this investigation you will examine a theory that chemists use to explain different aspects of chemical bonding valence shell electron pair repulsion vsepr theory attention will be given to how molecules are arranged in different shapes and how chemists can predict the geometry of a given molecule